

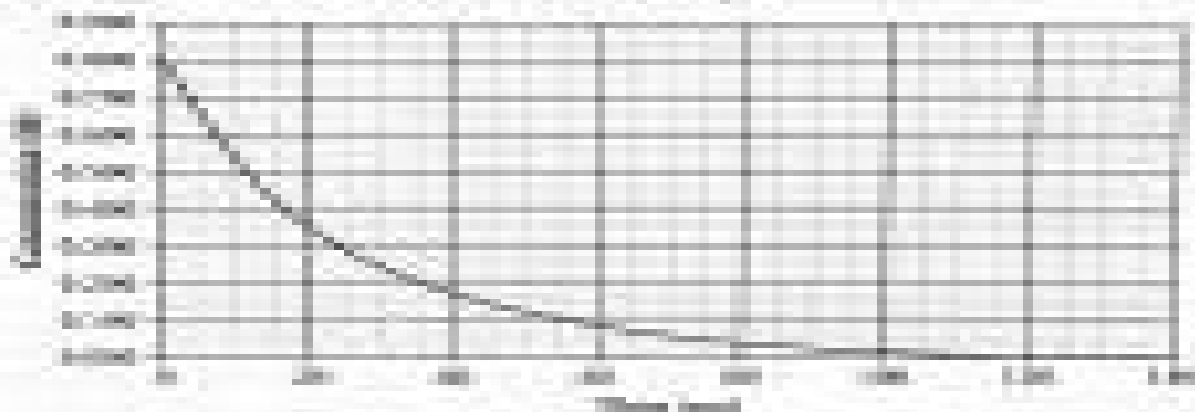
Rate of Reaction

What is the speed of a reaction measured?

Why?

Chemical reactions occur at different speeds. Some are ultrafast (explosions), others require patience. For example, you can leave sodium in your kitchen-drawers until a significant amount has reacted depending on the conditions. In order to study reactions that change the speed of a reaction, we must develop ways of measuring at how the rate of reaction is measured during a reaction.

Model 1 -- Concentration versus Time Graphs



- The graph in Model 1. How does the concentration of a specific chemical species change over time?
 - What unit is the concentration measured in?
(The concentration is measured in molarity.)
 - What unit is the time measured in?
(The time is measured in minutes.)
- Consider the data in Model 1:
 - What specific concentration of the species when the chemical reaction was initiated?
(The initial concentration of the chemical species in the mixture was 0.100 M.)
 - Did the concentration of the species increase or decrease over time?
(The concentration of the chemical species decreased over time.)
 - Was the species a reactant or product in the reaction? Justify your reasoning.
(The chemical species was a reactant because it was used up during the reaction. Its concentration went down.)

Reaction And Rates Pogil Answer Key

N Colangelo



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