

Chapter 11: Questions 1-10

Short answer: A long-answer question involving the system's behavior at equilibrium is assigned a maximum of 10 points.

Notes:

Objectives:

Objectives: Complete the following statements for your hypothesis. Consider one of the systems in brackets by highlighting, circling, or placing your answer in a different color.

If the concentration (mass) increases, the extent that [equilibrium constant] has been will [increase or decrease] to reach the new equilibrium.

If the concentration (mass) decreases, the extent that [equilibrium constant] will [increase or decrease] to reach the new equilibrium.

If the concentration (mass) increases, the extent that [equilibrium constant] will [increase or decrease] to reach the new equilibrium.

If the concentration (mass) decreases, the extent that [equilibrium constant] will [increase or decrease] to reach the new equilibrium.

Answer: The student's answer is not expected to be that long. The student's answer should be a short answer (only a few words) or a short answer (only a few words) or a short answer (only a few words).

Short answer: The student's answer is not expected to be that long.

System:	Short answer:
Equilibrium	
Equilibrium	
Equilibrium	

Objectives: Complete the following statements for your hypothesis. Consider one of the systems in brackets by highlighting, circling, or placing your answer in a different color.

Equilibrium constant (K) is:

$$K = \frac{[A][B]}{[C][D]}$$

Answer: The student's answer is not expected to be that long. The student's answer should be a short answer (only a few words) or a short answer (only a few words) or a short answer (only a few words).

Reaction Rates And Equilibrium Lab Report Answers

Albert A Gayle



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