#### Chapter 25 Nuclear Chemistry Practice Test

#### Multiple Choice

Identify the choice that best completes the statement or answers the question.

	1.	What	does	the 4	in	;He	represent?
--	----	------	------	-------	----	-----	------------

- a. the mass number
- b. the atomic number
- What does the 101 in <sup>256</sup><sub>101</sub>Md represent?
  - a. the mass number
  - b. the atomic number

- c. the nuclide number
- d. the number of neutrons

c. the number of protons

d. the number of neutrons

- b. 239 U

a. nuclear fusion.

c. radioactive decay.

e. electromagnetic waves.

b. nuclear radiation.

d. nuclear fission.

neutrons.

#### 6. Which of the following forms of radiation has the greatest penetrating power?

 a. alpha particles. b. beta particles.

c. gamma rays d. positrons

- Alpha particles are
  - a. electrons.
  - b. helium nuclei.
- 8. Beta particles are
  - a. electrons.
    - b. helium nuclei.

- electromagnetic waves. d. neutrons.
- 9. Which of the illustrations below represents a fission reaction?

# **Section 4 Review Nuclear Chemistry Answers**

Scott C. Dulebohn

## **Section 4 Review Nuclear Chemistry Answers:**

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