

Chapter 4 Atomic Structure

Section 4.1 Studying Atoms

(pages 100–105)

*This section discusses the development of atomic models.***Reading Strategy (page 100)**

Summarizing As you read, complete the table about atomic models. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Atomic Models		
Scientist	Evidence	Model
Dalton		Indivisible, solid spheres
	Deflected beam	Negative charges evenly scattered through positively charged mass of matter (plum pudding model)
	Deflection of alpha particles passing through gold foil	

Ancient Greek Models of Atoms (page 100)

- Democritus named the smallest particles of matter _____ because they could not be divided.

Dalton's Atomic Theory (page 101)

- Is the following sentence true or false? John Dalton gathered evidence for the existence of atoms by measuring the masses of elements that reacted to form compounds. _____
- Dalton's theory suggests that all matter is made up of individual particles called _____, which cannot be _____.
- Circle the letters of the sentences that represent main points of Dalton's theory of atoms.
 - All elements are composed of atoms.
 - In a particular compound, atoms of different elements always combine the same way.
 - All atoms have the same mass.

Section 4 1 Studying Atoms Answer Key

N Colangelo



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Vector Calculus Study Guide & Solutions Manual Karen Pao, Frederick Soon, 2003-08-22 Includes solutions to selected exercises and study hints CBSE Science Chapterwise Case Study Class 9 Priti Singhal, 2024-11-17 This book is structured to align with the latest syllabus and curriculum guidelines ensuring that the content is both relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging

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