

Chapter 10 Nuclear Chemistry

Section 10.2 Rates of Nuclear Decay**(pages 298–301)**

This section discusses half-lives and explains how nuclear decay can be used to estimate the age of objects.

Reading Strategy (page 298)

Identifying Details As you read, complete the concept map below to identify details about radiocarbon dating. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

**Half-life (pages 299–300)**

1. A nuclear decay rate describes how fast nuclear changes take place in a radioactive substance.
2. Is the following sentence true or false? All radioisotopes decay at the same rate. false
3. Describe a half-life. A half-life is the time required for one half of a sample of a radioisotope to decay.
4. Circle the letter that describes a sample of a radioisotope after two half-lives.
 - a. One eighth of the original sample is unchanged.
 - b. One quarter of the original sample is unchanged.**
 - c. Half of the original sample is unchanged.
 - d. Three quarters of the original sample is unchanged.
5. Circle the letter of the correct answer. Iodine-131 has a half-life of 8.07 days. What fraction of a sample of iodine-131 is left unchanged after 16.14 days?

a. $\frac{1}{2}$	b. $\frac{1}{4}$
c. $\frac{1}{8}$	d. $\frac{1}{16}$
6. Is the following sentence true or false? Like chemical reaction rates, nuclear decay rates vary with the conditions of reaction. false

Section 3 Rates Of Nuclear Decay Answers

Christian G. Meyer



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