

11 • Organic Chemistry

Organic Nomenclature Practice

Name / Functional Group	Formula	Structural Formula
n-butane	(1) C_4H_{10} (2) $CH_3CH_2CH_2CH_3$	
name & functional group 2-butanol alcohol name only	(1) and (2) C_4H_9OH $CH_3CH_2CH(OH)CH_3$	
trans-2-butene name only	(1) and (2) C_4H_8 $CH_3CH=CHCH_3$	
name only butyne	(1) and (2) C_4H_6 $CH_3C\equiv CCH_3$	
name only 1,3-butadiene	(1) and (2) C_4H_6 $CH_2=CHCH=CH_2$	
name only cyclopentane	(1) only C_5H_{10}	
name only cis-2-butene	(1) and (2) C_4H_8 $CH_3CH=CHCH_3$	
name only trans-2-butene	(1) and (2) C_4H_8 $CH_3CH=CHCH_3$	
name & functional group methylethyl ether	(1) and (2) C_3H_8O $CH_3CH_2OCH_3$	

Section Simple Organic Compounds Answers

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nonelectrolytes in a consistent manner After a description of the general characteristics of structure of solvents and solutions and the solubility of electrolytes and nonelectrolytes under normal temperature and pressure conditions the book first deals with different aspects of the density and the refractive index of solvents and dilute as well as concentrated solutions and finally with the transport i.e viscosity and electric conductivity and thermal properties of solvents and solutions Solvents and solutions is the first text devoted to the description and discussion of their properties since the publication of a monograph on the physical properties of aqueous electrolyte solutions more than three decades ago The main features of this book are Reflects developments in the investigation of solvents and solutions during the last three decades Outlines basic concepts involved in understanding the properties and behavior of solvents and solutions Describes and discusses different properties of ionic liquids as solvents and the behavior of their mixtures with other commonly used solvents Contents of different chapters are not only self contained but the contents are practically independent of each other Written as a practical guide for researchers who are looking for an uptodate overview of the physical and transport properties of solvents and solutions and as a reference source for workers in chemical industries and related fields and for graduate students of chemical engineering and physical chemistry

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