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alcohols ethanol 2 methylpropan-2-ol butan-2-ol but-1-en-1-ol $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ ii $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ iii $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ iv $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ aldehydes and ketones draw the structures for the following compounds ethanal butanone also known as butan-2-one but the 2 is unnecessary and is often left out 2-methylpropanal

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question 1 basic hydrocarbon nomenclature part a enter the number of carbons indicated by each of the following root names in the designated answer box part b the line formula for a branched alkane is shown below part c the line formula for another branched alkane is

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like putting too many or too few hydrogens on the middle carbon if you got it wrong re read the bit about naming alkanes and then look at these questions again ii

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dec 16 2021 so the complete name is 5 bromo 7 chloro 6 hydroxy 2 2 5 trimethyl 7 octen 4 one 4 it is not difficult to find the parent structure for this compound which is a cyclic alcohol so the last name is cyclopropanol the naming of the substituent with the benzene ring is

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solutions 1 aromatic compound iodobenzene 2 alkane 3 methylhexane 3 ketone 4 alkane alkyl halide 3 chloro 4 ethyl 2 4 dimethyloctane 5 aromatic compound o diethylbenzene or ortho diethylbenzene 6 alkene 3 methylpentene 7 ester 8 aromatic compound p bromotoluene or para bromotoluene 9 alkyne

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feb 21 2017 give the complete iupac name for each of the following compounds a b c give the complete iupac name and the common name for each of the following compounds a b c a 1 ethoxybutane butyl ethyl ether b 1 ethoxypropane ethyl propyl ether c 1 methoxypropane methyl propyl ether

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1 carbon: methane. 2 carbons: ethane. 3 carbons: propane. 4 carbons: butane. 5 carbons: pentane. 6 carbons: hexane. 7 carbons: heptane. 8 carbons: octane. 9 carbons: nonane. 10 carbons: decane. While many of these names share a Greek root with