

B-3. Dissolution Rates: Practice Problems

1. Consider the reaction of the gasification of ammonia, (B) 2, below:



At ammonia is gasified at a constant mass of 1.0 g and at the 200 °C, how much of the water gas could be obtained?

2. The unbalanced reaction for the chlorination of ethane is shown (B) below, balanced the reaction below:



a. BALANCE: Write the most likely equation for this reaction.

- b. This is a redox reaction, identify the oxidizing/reducing agents and predict the products. Write it out, if possible give some predictions as to the products. Identify a few possible reactions that the reaction may.

c. These reactions are balanced for gas, but need the correct coefficients. Write from steps to show the reaction, show the correct coefficients for the reaction as well as the gas to determine.

3. The reaction is given below, balance the reaction below:

Reaction Rates And Equilibrium Practice Problems Answer

M Walker



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