

Worksheet 1-1

- Measuring Reaction Rates

 Name - _____
 Due Date - _____

Answer the questions in the space provided. You must show all of your work to receive full marks. All answers MUST be rounded to the correct number of Sig. Figures.

1. A chemist wishes to determine the rate of reaction of zinc with hydrochloric acid. The equation for the reaction is:



A piece of zinc is dropped into 1.00 L of 0.100 M HCl and the following data were obtained:

Time	Mass of Zinc
0 s	0.016 g
4 s	0.014 g
8 s	0.012 g
12 s	0.010 g
16 s	0.008 g
20 s	0.006 g

$$\begin{array}{r} 0.016\text{g} \\ - 0.006\text{g} \\ \hline 0.010\text{g} \end{array}$$

- a) Calculate the **Rate of Reaction** in grams of Zn consumed per second.

$$\text{rate} = \frac{\Delta g}{\Delta \text{time}} = \frac{0.010\text{g}}{20\text{ sec}}$$

$$\text{Answer } 5.0 \times 10^{-4}$$

- b) Calculate the **Rate of Reaction** in moles of Zn consumed per second.

$$5.0 \times 10^{-4} \frac{\text{g}}{\text{sec}} \times \frac{1\text{mol}}{65.4\text{g}} =$$

$$\text{Answer } 7.6 \times 10^{-6} \frac{\text{mol}}{\text{sec}}$$

- c) Write out the complete ionic equation for the reaction.



- d) What will happen to the $[\text{H}^{+}]$ as the reaction proceeds? decrease

- e) What will happen to the $[\text{Cl}^{-}]$ as the reaction proceeds? nothing, it is a spectator

Reaction Rates Practice Problems Answers

**Patrick E. McMahon, Rosemary
McMahon, Bohdan Khomtchouk**



Reaction Rates Practice Problems Answers:

(Free Sample) *GO TO Objective NEET Chemistry Guide with DPP & CPP Sheets 9th Edition* Disha Experts, 2021-10-07

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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13. This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific detailed step-by-step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid-base reactions, which often persists through working with acid-base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields.

Chemistry Rob Ritchie, 2004. These New editions of the successful highly illustrated study revision guides have been fully updated to meet the latest specification changes. Written by experienced examiners, they contain in-depth coverage of the key information plus hints, tips, and guidance about how to achieve top grades in the A2 exams. Progress check questions, test recall and understanding, and end of unit sample questions and model answers provide essential practice to improve students' exam technique.

GO TO Objective NEET 2021 Chemistry Guide 8th Edition Disha Experts, **General Organic and Biological Chemistry** Kenneth W. Raymond, 2009-12-14. This general organic and biochemistry text has been written for students preparing for careers in health-related fields such as nursing, dental hygiene, nutrition, medical technology, and occupational therapy. It is

also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry. Students need have no previous background in chemistry but should possess basic math skills. The text features numerous helpful problems and learning features.

Partial Solutions Manual Darrell D. Ebbing, David Bookin, 1998

Survival Guide to Organic Chemistry Patrick E. McMahon, Bohdan B. Khomtchouk, Claes Wahlestedt, 2016-12-19

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Organic Chemistry T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-01-19 The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder s tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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Introduction to the Pharmaceutical Sciences Nita K. Pandit, 2007 This unique textbook provides an introductory yet

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