

5.1 Atomic structure and the Periodic Table

You will learn:

- To understand how the position of an element in the Periodic Table can be used to predict its atomic structure and chemical and physical properties.
- To understand that models and analogies can change according to scientific evidence
- To describe the uses, strengths and limitations of models and analogies
- To represent scientific ideas using recognised symbols or formulae

1.

Draw a line to match the word with its correct meaning.

The first line has been drawn for you.

Word

Meaning

Electrons

Particles with no charge found in an atom

Neutrons

The central part of the atom

Protons

The paths or orbits that electrons move along in the atom

Nucleus

Very small negatively charged particles in an atom

Shells

Particles with a positive charge found in an atom

2.

Complete the sentences.

Use the words in the box.

atomic

mass

protons

electrons

nucleus

shells

neutrons

atom

a An _____ number is the protons an _____ of an element has.

b A _____ number is the total number of _____ and _____ found in the nucleus of an atom.

Periodic Table Workbook Answers Chemistry

Michael Brown



Periodic Table Workbook Answers Chemistry:

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