

Periodic Table of the Elements

1 IA H Hydrogen (1.008)																	18 VIIIA He Helium (4.003)		
3 IIA Li Lithium (6.941)	4 IIA Be Beryllium (9.012)													5 IIIA B Boron (10.81)	6 IIIA C Carbon (12.011)	7 IVA N Nitrogen (14.007)	8 IVA O Oxygen (15.999)	9 VA F Fluorine (18.998)	10 VA Ne Neon (20.180)
11 IA Na Sodium (22.990)	12 IIA Mg Magnesium (24.305)													13 IIIA Al Aluminum (26.982)	14 IIIA Si Silicon (28.086)	15 IVA P Phosphorus (30.974)	16 IVA S Sulfur (32.06)	17 VA Cl Chlorine (35.45)	18 VA Ar Argon (39.948)
19 IA K Potassium (39.098)	20 IIA Ca Calcium (40.078)	21 IIB Sc Scandium (44.956)	22 IIB Ti Titanium (47.88)	23 IIB V Vanadium (50.942)	24 IIB Cr Chromium (51.996)	25 IIB Mn Manganese (54.938)	26 IIB Fe Iron (55.845)	27 IIB Co Cobalt (58.933)	28 IIB Ni Nickel (58.693)	29 IB Cu Copper (63.546)	30 IB Zn Zinc (65.38)	31 IIB Ga Gallium (69.723)	32 IIB Ge Germanium (72.63)	33 IIIA As Arsenic (74.922)	34 IIIA Se Selenium (78.96)	35 IIIA Br Bromine (79.904)	36 IIIA Kr Krypton (83.798)		
37 IA Rb Rubidium (85.468)	38 IIA Sr Strontium (87.62)	39 IIB Y Yttrium (88.906)	40 IIB Zr Zirconium (91.224)	41 IIB Nb Niobium (92.906)	42 IIB Mo Molybdenum (95.94)	43 IIB Tc Technetium (98.906)	44 IIB Ru Ruthenium (101.07)	45 IIB Rh Rhodium (102.91)	46 IIB Pd Palladium (106.36)	47 IB Ag Silver (107.865)	48 IB Cd Cadmium (112.411)	49 IIB In Indium (114.818)	50 IIB Sn Tin (118.710)	51 IIIA Sb Antimony (121.757)	52 IIIA Te Tellurium (127.6)	53 IIIA I Iodine (126.905)	54 IIIA Xe Xenon (131.29)		
55 IA Cs Cesium (132.905)	56 IIA Ba Barium (137.327)	57-71 Lanthanides	72 IIB Hf Hafnium (178.49)	73 IIB Ta Tantalum (180.948)	74 IIB W Tungsten (183.84)	75 IIB Re Rhenium (186.207)	76 IIB Os Osmium (190.23)	77 IIB Ir Iridium (192.222)	78 IIB Pt Platinum (195.084)	79 IB Au Gold (196.967)	80 IB Hg Mercury (200.59)	81 IIB Tl Thallium (204.383)	82 IIB Pb Lead (207.2)	83 IIIA Bi Bismuth (208.98)	84 IIIA Po Polonium (209)	85 IIIA At Astatine (210)	86 IIIA Rn Radon (222)		
87 IA Fr Francium (223)	88 IIA Ra Radium (226)	89-103 Actinides	104 IIB Rf Rutherfordium (261)	105 IIB Db Dubnium (262)	106 IIB Sg Seaborgium (266)	107 IIB Bh Bohrium (264)	108 IIB Hs Hassium (277)	109 IIB Mt Meitnerium (268)	110 IIB Ds Darmstadtium (271)	111 IB Rg Roentgenium (272)	112 IB Cn Copernicium (285)	113 IIB Nh Nihonium (284)	114 IIB Fl Flerovium (289)	115 IIIA Mc Moscovium (288)	116 IIIA Lv Livermorium (293)	117 IIIA Ts Tennessine (289)	118 IIIA Og Oganesson (294)		

Atomic Number →

→ Symbol

→ Name

→ Atomic Weight

→ Electrons per shell

1

H

Hydrogen

1.008

1

57 La Lanthanum (138.905)	58 Ce Cerium (140.12)	59 Pr Praseodymium (140.908)	60 Nd Neodymium (144.24)	61 Pm Promethium (145)	62 Sm Samarium (150.36)	63 Eu Europium (151.964)	64 Gd Gadolinium (157.25)	65 Tb Terbium (158.925)	66 Dy Dysprosium (162.50)	67 Ho Holmium (164.930)	68 Er Erbium (167.259)	69 Tm Thulium (168.933)	70 Yb Ytterbium (173.054)	71 Lu Lutetium (174.967)
89 Ac Actinium (227)	90 Th Thorium (232.038)	91 Pa Protactinium (231.036)	92 U Uranium (238.029)	93 Np Neptunium (237.048)	94 Pu Plutonium (244.064)	95 Am Americium (243.061)	96 Cm Curium (247.07)	97 Bk Berkelium (247.07)	98 Cf Californium (251.08)	99 Es Einsteinium (252.083)	100 Fm Fermium (257.10)	101 Md Mendelevium (258.10)	102 No Nobelium (259.10)	103 Lr Lawrencium (262.10)

Periodic Table Section 3

Yan Bai



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2023-24)Onward V. K. Sally,2023-05-20 A New Approach to ICSE Chemistry for Class X has been revised in accordance with the latest syllabus prescribed by the Council for the Indian School Certificate Examination New Delhi for Class X The main strength of the book lies in the presentation of scientific content which has been so arranged that the topics are linked with each other and do not cause any unnecessary burden on the mind of the student Salient features of this book are as follows Clear simple and easy language A large number of chemical reactions along with experimental observations Full sized diagrams as expected from a student in the examination Topic wise Video Lectures as a support for effective learning Concepts supplemented by suitable day to day examples Periodic Table showing mass number atomic number of various elements along with the list of actual names of the elements Highlighted important terms and definitions Well designed exercises to assess conceptual reasoning skill and application based learning as per the latest recommendations of ICSE board Solved numerical problems at the end of each chapter to help the students solve numericals on their own A chapter on Practical Chemistry to help students in their laboratory work The latest Solved ICSE Specimen Paper has been given Scan QR codes given at the end of each chapter to get the solution of chapter wise ICSE Board Examination Questions We hope this book will prove useful to fellow teachers and students Suggestions for further improvement of this book shall be gratefully acknowledged Author Super 10 CBSE Class 12 Chemistry 2021-22 Term I Sample Papers with OMR Sheets Disha Experts,2021-09-01

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