

Periodic Trends Practice Worksheet

Use the periodic table and your knowledge of the periodic trends to answer the following questions.

- ① Identify each element as a metal, metalloid, or nonmetal
 - Ⓐ germanium _____
 - Ⓑ zinc _____
 - Ⓒ phosphorous _____
 - Ⓓ lithium _____
- ② Which of the two species is larger in each of the following pairs?
 - Ⓐ N^{3-} or F^- _____
 - Ⓑ Mg^{2+} or Ca^{2+} _____
 - Ⓒ Fe^{2+} or Fe^{3+} _____
- ③ Which of the two species is smaller in each of the following pairs?
 - Ⓐ K^+ or Li^+ _____
 - Ⓑ Au^+ or Au^{3+} _____
 - Ⓒ P^{3-} or N^{3-} _____
- ④ What atom has the LARGEST atomic radius?

- ⑤ What atom has the SMALLEST atomic radius?

- ⑥ Which is larger?
 - Ⓐ Mg or Mg^{2+} _____
 - Ⓑ I or I^- _____
 - Ⓒ Cl^- or Br^- _____
 - Ⓓ F^- or Ne _____
 - Ⓔ N^{3-} or P^{3-} _____
- ⑦ Put the following in order of increasing radius: Br^- Se^{2-} Sr^{2+} Kr Rb^+
- ⑧ Circle the correct answer.
 - Ⓐ Bromine has a higher / lower ionization energy than Lead
 - Ⓑ Barium has a higher / lower ionization energy than Radon
 - Ⓒ Fluorine has a higher / lower ionization energy than Arsenic
 - Ⓓ Lithium has a higher / lower ionization energy than Cesium
- ⑨ Which of the two species is larger in each of the following pairs?
 - Ⓐ Mg or Sr
 - Ⓑ Be or N
 - Ⓒ P or Cl
 - Ⓓ Ni or Br
- ⑩ Arrange the following in order of increasing electronegativity: S , O , Ne , Al

Periodic Trends Practice Problems With Answers

Disha Experts

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the 'Disha Experts' text.

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is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner

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