

CHAPTER 5: PERIODICITY

The Periodic Law

SECOND PERIOD

Student's Assistant Answered the following questions for the second period.

1. (a) Consider the halogen atoms with 10 protons, and the elements adjacent to the halogen group.

- _____ (a) The first is the atomic number?
- _____ (b) The first is the atomic number?
- _____ (c) Is the element is periodic in a regular periodic table observed by the periodic table? or by the periodic table?

2. (a) Consider the elements with the following electronic configurations: $1s^2 2s^2 2p^2$, $1s^2 2s^2 2p^3$.

- _____ (a) The first is the atomic number?
- _____ (b) The first is the atomic number?
- _____ (c) The group number is equal to $10 + 2 = 12$. That is, the first.

- _____ (d) The first is the atomic number?
- _____ (e) The first is the atomic number?

- _____ (f) The first is the atomic number?

- _____ (g) The first is the atomic number?
- _____ (h) The first is the atomic number?

3. (a) Consider the periodic table. In the top of the periodic table, the elements are arranged in the following order: $1s^2 2s^2 2p^2$, $1s^2 2s^2 2p^3$.

- _____ (a) The first is the atomic number?
- _____ (b) The first is the atomic number?
- _____ (c) The first is the atomic number?
- _____ (d) The first is the atomic number?
- _____ (e) The first is the atomic number?
- _____ (f) The first is the atomic number?
- _____ (g) The first is the atomic number?
- _____ (h) The first is the atomic number?
- _____ (i) The first is the atomic number?
- _____ (j) The first is the atomic number?
- _____ (k) The first is the atomic number?
- _____ (l) The first is the atomic number?
- _____ (m) The first is the atomic number?
- _____ (n) The first is the atomic number?
- _____ (o) The first is the atomic number?
- _____ (p) The first is the atomic number?
- _____ (q) The first is the atomic number?
- _____ (r) The first is the atomic number?
- _____ (s) The first is the atomic number?
- _____ (t) The first is the atomic number?
- _____ (u) The first is the atomic number?
- _____ (v) The first is the atomic number?
- _____ (w) The first is the atomic number?
- _____ (x) The first is the atomic number?
- _____ (y) The first is the atomic number?
- _____ (z) The first is the atomic number?

Periodic Law Mixed Review Answer Key

Richard Bailey

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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