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**User's Guide**

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c and the change in temperature is 50 c 20 c 30 c plugging these values into the equation we get  $q = 75.0 \text{ g} \times 1 \text{ cal/g} \times 30 \text{ }^{\circ}\text{C} = 2250$   
cal therefore 2250 cal is the correct answer rate this question 2 2

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