# Personal Finance Chapter 2 Exam Questions with Correct Answers

Money management refers to

- A. Preparing personal financial statements.
- B. Day-to-day financial activities.
- C. Trade-offs that occur with financial decisions.
- D. Storing financial records for easy access.
- E. Spending money on current living expenses. Answer-b

Which of the following is NOT a component of money management?

- A. Storing personal financial records to document business transactions and legal matters.
- B. Creating personal financial statements to measure and assess financial position and progress.
- C. Creating a budget.
- D. None of these are components of money management.

F

Storing personal financial records, creating personal financial statements, and creating a budget are all

components of money management. - Answer-e

A home file should be used for storing

- A. All financial documents and records.
- B. Financial records for current needs.
- C. Documents that require maximum security.
- D. Obsolete financial documents.
- E. Records that are difficult to replace. Answer-b

Which of the following financial documents would most likely be stored in a safe deposit box?

- A. W-2 forms
- B. Personal financial statements
- C. Warranties
- D. Marriage certificates
- E. Checking account statements Answer-d

# **Personal Finance Chapter Quiz**

**Harold Arthur Wolf** 

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web the lanthanide 'læn $\theta$ ənaid or lanthanoid 'læn $\theta$ ənaid series of chemical elements a comprises the 14 metallic chemical elements with atomic numbers 57 70 from lanthanum through ytterbium 2 3 4 lutetium is also sometimes considered a lanthanide despite being a d block element and a transition metal

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web five of the elements la ce pr nd eu are very reactive and when exposed to air react with oxygen to form an oxide coating that tarnishes the surface for this reason these metals are stored under mineral oil the remainder of the lanthanides are not as reactive and some gd lu retain their silvery metallic appearance for a long time

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web to separate the lanthanides from other elements occurring with them they are chemically combined with specific substances to form lanthanide compounds with low solubility oxalates and fluorides for example a process known as ion exchange is then used to separate the lanthanides from each other in this process a solution of the

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web the fifteen elements shown in table pageindex 1 from lanthanum la 4f 0 to lutetium lu 4f 14 are lanthanoids ln may be used as a general symbol for the lanthanoid elements although lanthanoids scandium sc and yttrium y are sometimes called rare earth elements they are relatively abundant in the earth s crust

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web feb 4 2021 the lanthanides are located on the top row of the two rows of elements found below the main body of the periodic table placing the lanthanides and the actinides the row below the lanthanides below the other elements is just a matter of aesthetics the lanthanides belong to period 6 like cesium barium and hafnium