

PERIODIC TABLE (ADVANCED)

Molecular Structure of Hydrogen



Atomic symbol: H
Atomic number: 1
Atomic weight: 1.0079

Basic Information

Name: Hydrogen

Symbol: H

Atomic Number: 1

Atomic Mass: 1.00794 amu

Melting Point: -259.14 °C (14.009985 K, -434.45203 °F)

Boiling Point: -252.87 °C (20.280005 K, -423.166 °F)

Number of Protons/Electrons: 1

Number of Neutrons: 0

Classification: Non-metal

Crystal Structure: Hexagonal

Density @ 293 K: 0.08988 g/cm³

Color: colorless

Atomic Structure

Number of Energy Levels: 1

First Energy Level: 1

Isotopes

Isotope	Half Life
H-1	Stable
H-2	Stable
H-3	12.3 years

Facts

Name Origin: Greek words *hudeir* (water) and *gennan* (generate)

Uses: Balloons, metal refining

Obtained From: mines, oil, gas wells

Molecular Structure of Carbon



Atomic symbol: C
Atomic number: 6
Atomic weight: 12.011

Name: Carbon

Symbol: C

Atomic Number: 6

Atomic Mass: 12.0107 amu

Melting Point: 3500.0 °C (3773.15 K, 6332.0 °F)

Boiling Point: 4827.0 °C (5100.15 K, 8720.6 °F)

Number of Protons/Electrons: 6

Number of Neutrons: 6

Classification: Non-metal

Crystal Structure: Hexagonal

Density @ 293 K: 2.62 g/cm³

Color: May be black

Atomic Structure

Number of Energy Levels: 2

First Energy Level: 2 Second Energy Level: 4

Isotopes

Isotope	Half Life
C-11	20.3 minutes
C-12	Stable
C-13	Stable
C-14	5730.0 years
C-15	2.5 seconds

Facts

Uses: steel, filters

Obtained From: burning with insufficient oxygen

Periodic Table Advanced Speedy Study Guide

Jicheng Xie



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Table of Contents Periodic Table Advanced Speedy Study Guide

1. Understanding the eBook Periodic Table Advanced Speedy Study Guide
 - The Rise of Digital Reading Periodic Table Advanced Speedy Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Advanced Speedy Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Periodic Table Advanced Speedy Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Advanced Speedy Study Guide
 - Personalized Recommendations
 - Periodic Table Advanced Speedy Study Guide User Reviews and Ratings
 - Periodic Table Advanced Speedy Study Guide and Bestseller Lists

5. Accessing Periodic Table Advanced Speedy Study Guide Free and Paid eBooks
 - Periodic Table Advanced Speedy Study Guide Public Domain eBooks
 - Periodic Table Advanced Speedy Study Guide eBook Subscription Services
 - Periodic Table Advanced Speedy Study Guide Budget-Friendly Options
6. Navigating Periodic Table Advanced Speedy Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Advanced Speedy Study Guide Compatibility with Devices
 - Periodic Table Advanced Speedy Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Advanced Speedy Study Guide
 - Highlighting and Note-Taking Periodic Table Advanced Speedy Study Guide
 - Interactive Elements Periodic Table Advanced Speedy Study Guide
8. Staying Engaged with Periodic Table Advanced Speedy Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Advanced Speedy Study Guide
9. Balancing eBooks and Physical Books Periodic Table Advanced Speedy Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Advanced Speedy Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Advanced Speedy Study Guide
 - Setting Reading Goals Periodic Table Advanced Speedy Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Advanced Speedy Study Guide
 - Fact-Checking eBook Content of Periodic Table Advanced Speedy Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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