

Trends on the Periodic Table

Part 1: Identifying chemical properties/characteristics of the elements

For each of the following elements, draw a caricature or cartoon above the description to show its "chemical personality" based on the description provided. (Note: See the Neon example for ideas.)

Lithium
Being one of the smaller elements with only 3 electrons, lithium is happy to give one of its electrons away and never seems to ask for any in return.



Fluorine
One of the smallest elements, Fluorine is often described as snappy with a "magnetic" personality. As part of the halogen family (stoutest electron thieves), Fluorine is the best at convincing other elements to give up their electrons to him. However, he holds on tight to his own electrons and will not let them go without a fight.

I am so perfect... I have a full electron shell. I don't need those other elements... I hope they just leave me alone!



Neon
Part of the Noble family, Neon does not readily interact with other elements. With a full set of valence electrons, Neon feels that he has everything he needs to make him happy without sharing with others.

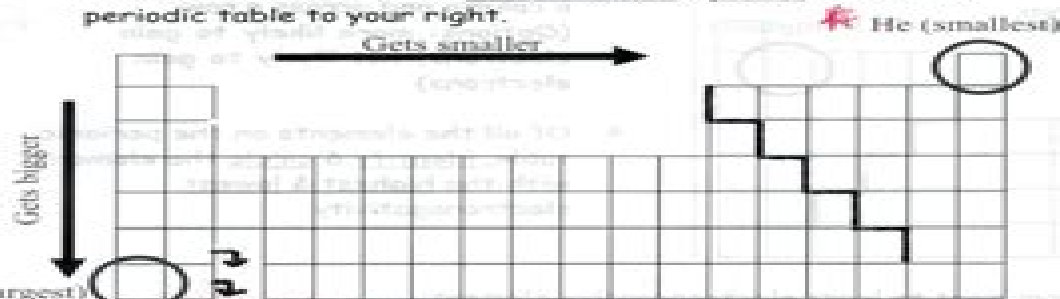
Francium
Francium is the largest of all of the elements and has so many electrons that it has trouble keeping track of them. Francium will eagerly give up electrons to other elements and never takes in extras. But watch out, it has an extremely reactive and explosive personality!

Iodine
As part of the halogen family, Iodine is also a skilled electron thief, however, not nearly as good as its smaller brothers and sisters. Iodine will also put up a pretty good fight to keep its own electrons.

Part 2: Use your textbook to complete the following:

Atomic Radii Size:

1. Explain **atomic radius**: A measure of the size (volume) of an atom. Determined by measuring half of the distance between two identical atoms bonded together.
2. Using the information in figure 5-13 (pg 141) in your book, **write** the atomic radii size for **alkali metals & period #2** on the periodic table to your right.



3. What happens to atomic radii as you move down a column and across a row? On the arrows on the diagram, indicate the trend. (Options: gets bigger OR gets smaller)
4. Of all the elements on the periodic table, **identify & circle the largest & smallest** in size.

5. Rank the following elements from largest to smallest in atomic radii:

a) Be, Ba, Mg = **Ba, Mg, Be**

b) Li, F, B = **Li, B, F**

6. How can you explain the trend?

a. Why do they get smaller as you go across a period? Additional protons in the nucleus "pull" on electrons, bring the electron cloud closer.

Effective nuclear charge, Z_{eff} , increases the attraction of the nucleus and therefore pulls the electron cloud closer to the nucleus resulting in smaller radii

not a good enough answer
4AP Answer

Periodic Trends Worksheet Pogil Extension Questions

Raffaela Di Napoli



Periodic Trends Worksheet Pogil Extension Questions:

The book delves into Periodic Trends Worksheet Pogil Extension Questions. Periodic Trends Worksheet Pogil Extension Questions is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Periodic Trends Worksheet Pogil Extension Questions, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Periodic Trends Worksheet Pogil Extension Questions
 - Chapter 2: Essential Elements of Periodic Trends Worksheet Pogil Extension Questions
 - Chapter 3: Periodic Trends Worksheet Pogil Extension Questions in Everyday Life
 - Chapter 4: Periodic Trends Worksheet Pogil Extension Questions in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Periodic Trends Worksheet Pogil Extension Questions. The first chapter will explore what Periodic Trends Worksheet Pogil Extension Questions is, why Periodic Trends Worksheet Pogil Extension Questions is vital, and how to effectively learn about Periodic Trends Worksheet Pogil Extension Questions.
 3. In chapter 2, the author will delve into the foundational concepts of Periodic Trends Worksheet Pogil Extension Questions. This chapter will elucidate the essential principles that must be understood to grasp Periodic Trends Worksheet Pogil Extension Questions in its entirety.
 4. In chapter 3, this book will examine the practical applications of Periodic Trends Worksheet Pogil Extension Questions in daily life. The third chapter will showcase real-world examples of how Periodic Trends Worksheet Pogil Extension Questions can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Periodic Trends Worksheet Pogil Extension Questions in specific contexts. The fourth chapter will explore how Periodic Trends Worksheet Pogil Extension Questions is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Periodic Trends Worksheet Pogil Extension Questions. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Periodic Trends Worksheet Pogil Extension Questions.

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