# CHEMISTRY SEMESTER B: FINAL EXAM REVIEW EXAM QUESTIONS WITH 100% CORRET ANSWERS ALREADY GRADED A+

What atoms follow the Duet role? - answer Li and Na.

What type of bonding takes place between N and F? - answer covalent

How many electrons is an octet? - answer 8

To draw the Lewis Structure of the polyatomic ion CIO2, you would have to \_\_\_\_\_\_ those in the structures of CI, O, O and O. - answer add one electron to

The correct Lewis structure for a molecule of the compound C2H2 contains - answer one triple bond

Which of the following atoms has six valence electrons? - answer Sulfur

What is this structure for H2? - answer H-H

In a double bond, two electrons share a total of \_\_\_\_\_\_ electrons, - answer four

How many moles of HCl are present in 0.70L of a 0.33 M HCl solution - answer 0.23mol

What is the molarity of a solution that contains 125 g NaCl in 4.00 L solution? - answer 0.535M

How much KCI can you dissolve in 200g of 40 C water according to figure 1? - answer 80g.

# **Plato Course Chemistry Semester B Answers**

**Victor M. Corman** 

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