

**Table K  
Selected Polyatomic Ions**

Formula	Name	Formula	Name
$\text{ClO}_4^-$	perchlorate	$\text{ClO}_2^-$	chlorite
$\text{ClO}_3^-$	chlorate	$\text{ClO}_2^+$	dichlorine oxide
$\text{NH}_4^+$	ammonium	$\text{MnO}_4^-$	permanganate
$\text{C}_2\text{H}_3\text{O}_2^-$ $\text{CH}_3\text{COO}^-$	acetate	$\text{NO}_2^-$	nitrite
$\text{CN}^-$	cyanide	$\text{NO}_3^-$	nitrate
$\text{CO}_3^{2-}$	carbonate	$\text{O}_2^{2-}$	peroxide
$\text{HCO}_3^-$	hydrogen carbonate	$\text{OH}^-$	hydroxide
$\text{C}_2\text{O}_4^{2-}$	oxalate	$\text{PO}_4^{3-}$	phosphate
$\text{F}_3\text{B}^-$	superfluoride	$\text{SCN}^-$	thiocyanate
$\text{F}_3\text{C}^-$	trifluoride	$\text{SO}_3^{2-}$	sulfite
$\text{IO}_3^-$	iodate	$\text{SO}_4^{2-}$	sulfate
$\text{IO}_4^-$	periodate	$\text{HSO}_4^-$	hydrogen sulfate
$\text{ClO}_2^+$	chlorine dioxide	$\text{S}_2\text{O}_8^{2-}$	persulfate

**Table L  
Solubility Guidelines for Aqueous Solutions**

Basic Most Forms Soluble Compounds	Exceptions	Basic Most Forms Soluble Compounds*	Exceptions
Group 1 ions ( $\text{Li}^+$ , $\text{Na}^+$ , etc.)		acetate ( $\text{CH}_3\text{CO}_2^-$ )	when coordinated with Group 1 ions or ammonium ( $\text{NH}_4^+$ )
ammonium ( $\text{NH}_4^+$ )		phosphate ( $\text{PO}_4^{3-}$ )	when coordinated with Group 1 ions, $\text{NH}_4^+$ , or ammonium ( $\text{NH}_4^+$ )
nitrate ( $\text{NO}_3^-$ )		carbonate ( $\text{CO}_3^{2-}$ )	when coordinated with Group 1 ions or ammonium ( $\text{NH}_4^+$ )
acetate ( $\text{CH}_3\text{CO}_2^-$ )		hydroxide ( $\text{OH}^-$ )	when coordinated with Group 1 ions, $\text{NH}_4^+$ , or ammonium ( $\text{NH}_4^+$ )
hydroxide ( $\text{OH}^-$ )		halides ( $\text{Cl}^-$ , $\text{Br}^-$ , $\text{I}^-$ )	when coordinated with $\text{Ag}^+$ , $\text{Cu}^+$ , or $\text{Hg}_2^{2+}$
halides ( $\text{Cl}^-$ , $\text{Br}^-$ , $\text{I}^-$ )		sulfates ( $\text{SO}_4^{2-}$ )	when coordinated with $\text{Ag}^+$ , $\text{Ca}^{2+}$ , or $\text{Sr}^{2+}$
sulfates ( $\text{SO}_4^{2-}$ )			

\*Ammonium is having very low solubility in  $\text{H}_2\text{O}$

# New York State Ratings Guide Chemistry Regents

**AW Chickering**



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## **Table of Contents New York State Ratings Guide Chemistry Regents**

1. Understanding the eBook New York State Ratings Guide Chemistry Regents
  - The Rise of Digital Reading New York State Ratings Guide Chemistry Regents
  - Advantages of eBooks Over Traditional Books
2. Identifying New York State Ratings Guide Chemistry Regents
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in a New York State Ratings Guide Chemistry Regents
  - User-Friendly Interface
4. Exploring eBook Recommendations from New York State Ratings Guide Chemistry Regents
  - Personalized Recommendations
  - New York State Ratings Guide Chemistry Regents User Reviews and Ratings

- New York State Ratings Guide Chemistry Regents and Bestseller Lists
- 5. Accessing New York State Ratings Guide Chemistry Regents Free and Paid eBooks
  - New York State Ratings Guide Chemistry Regents Public Domain eBooks
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  - New York State Ratings Guide Chemistry Regents Budget-Friendly Options
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  - New York State Ratings Guide Chemistry Regents Compatibility with Devices
  - New York State Ratings Guide Chemistry Regents Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of New York State Ratings Guide Chemistry Regents
  - Highlighting and Note-Taking New York State Ratings Guide Chemistry Regents
  - Interactive Elements New York State Ratings Guide Chemistry Regents
- 8. Staying Engaged with New York State Ratings Guide Chemistry Regents
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers New York State Ratings Guide Chemistry Regents
- 9. Balancing eBooks and Physical Books New York State Ratings Guide Chemistry Regents
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection New York State Ratings Guide Chemistry Regents
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine New York State Ratings Guide Chemistry Regents
  - Setting Reading Goals New York State Ratings Guide Chemistry Regents
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of New York State Ratings Guide Chemistry Regents
  - Fact-Checking eBook Content of New York State Ratings Guide Chemistry Regents
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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