



Name _____ Class _____ Date _____

1

How many atoms of **oxygen (O)** are in the **chemical formula** below?

- A 4
- B 5
- C 7
- D 12

**3**

Carbon (C) and hydrogen (H) create **covalent bonds** to share **electrons** between them. What is the total number of electrons being shared between **carbon** and **hydrogen** in the molecule of propane shown below?

- A 8
- B 11
- C 16
- D 32

**2**

The diagram shown below is a **structural formula** for the gas **propane**. What would be the correct **chemical formula** for propane?

- A C_3H_6
- B CH_6
- C $8\text{H}_3\text{C}$
- D CH

**4**

The **reason** why the bonds between carbon (C) and hydrogen (H) are **covalent** is _____

- A C and H do not have electrons to give away
- B C and H are both nonmetals
- C C and H are both metalloids
- D C and H can bond only by sharing electrons

**5**

The diagram below represents the **structural formula** for several **water molecules**. How many **molecules** are represented?

- A 3
- B 5
- C 10
- D 15

**6**

Explain the **arrangement** of the molecules in the diagram of water below.

- A it is a lucky arrangement
- B water molecules are sticky
- C water molecules have polarity
- D water molecules are magnetic

**7**

Water molecules have some **polarity** in their structure. Using the diagram below, notice how the ends of H atoms (**positive**) and O atoms (**negative**) meet. This property is the **reason** that _____

- A water forms droplets easily
- B water boils easily
- C ice melts easily
- D water has color

**8**

Since H atoms are slightly **positive** and the O atoms are slightly **negative**, molecules of water have **polarity**. This makes water a **good** _____

- A solute
- B refrigerant
- C solvent
- D magnet

**9**

The chemical formula for the **oxygen** that we breathe is _____

- A O
- B O_2
- C O_3
- D 2O

**10**

How many **molecules of oxygen** are represented in the chemical formula below?

- A 2
- B 4
- C 6
- D 18



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Capturing 21st century skills UNESCO Office Dakar and Regional Bureau for Education in Africa,Kim, Helyn,Care, Esther,2020-08-20

Algebra 2 Made Easy Common Core Edition MaryAnn Casey Topical Review Book Company,2015-12-31 A quick how to do it reference guide recommended for students studying the new Common Core standards for Algebra 2 Loaded with lots of examples definitions and correlations

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Research Report Kentucky. General Assembly. Legislative Research Commission,1960

National Weather Service Staff Reductions Proposed for Fiscal Year 1997 and Projected for Fiscal Year 1998 United States. Congress. Senate. Committee on Commerce, Science, and Transportation. Subcommittee on Science, Technology, and Space,1998

8th Grade OAA Test Prep Study Guide Mark Shannon,2018-01-18 This test prep is good for kids who are on the fence and having trouble passing or need to increase their margin These kids will often say I DON T KNOW WHERE TO START when explaining their confusion This book address that problem with three approaches 1 In contrast with the other study guides

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Canadian Periodical Index , 1996 *Cumulative Author Index to Psychological Abstracts* , 1972 *Spectrum Test Prep, Grade 8* Spectrum, 2015-01-05 Spectrum Test Prep Grade 8 includes strategy based activities for language arts and math test tips to help answer questions and critical thinking and reasoning The Spectrum Test Prep series for grades 1 to 8 was developed by experts in education and was created to help students improve and strengthen their test taking skills The

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