

Level		Skills descriptors
Level 4 19–24 marks Compelling, Convincing Communication	Upper Level 4 22–24 marks	Content • Communication is convincing and compelling • Tone, style and register are assuredly matched to purpose and audience • Extensive and ambitious vocabulary with sustained crafting of linguistic devices Organisation • Varied and inventive use of structural features • Writing is compelling, incorporating a range of convincing and complex ideas • Fluently linked paragraphs with seamlessly integrated discourse markers
	Lower Level 4 19–21 marks	Content • Communication is convincing • Tone, style and register are convincingly matched to purpose and audience • Extensive vocabulary with conscious crafting of linguistic devices Organisation • Varied and effective structural features • Writing is highly engaging with a range of developed complex ideas • Consistently coherent use of paragraphs with integrated discourse markers

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Annelies Wilder-Smith



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Whispering the Strategies of Language: An Emotional Journey through **Ppe Paper May 2015 Mark Scheme**

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide Ca(OH)_2 by measuring the

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of Ca(OH)_2 is represented in equation form as shown below $\text{Ca(OH)}_2(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq})$ the solubility product expression describes in mathematical terms the equilibrium

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web the chemical equation used was $\text{Ca(OH)}_2(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq})$ the solubility product was determined to be ksp $\text{Ca}^{2+}\text{OH}^{-}$ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error
accepted value for K_{sp} for sodium hydroxide is 5×10^{-6}

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