

# Quiz & Worksheet - Determining the Chemical Formula of Hydrates from Empirical Data

#### 1. What is the difference between a hydrate and an anhydrate?

- An anhydrate is a hydrate with the water removed.
- A hydrate is an anhydrate with the water removed.
- A hydrate is necessary for producing water molecules, but an anhydrate isn't.
- An anhydrate is necessary for producing water molecules, but a hydrate isn't.

#### 2. In the formula MgSO<sub>4</sub> nH<sub>2</sub> O, what does the n before the H<sub>2</sub> O stand for?

- The number of moles of water.
- The number of moles of hydrogen.
- The number of moles of hydrate.
- The number of moles of anhydrate.

#### 3. If there are three moles of H<sub>2</sub> O for every one mole of LiClO4, then what is the hydrate formula?

- o LICIO<sub>4</sub>, 3H<sub>2</sub> O
- 3LICIO<sub>4</sub>: 3H<sub>2</sub> O
- o LICIO4: 0H2 003
- 3(LICIO<sub>4</sub>): 3H<sub>2</sub> O

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