NUCLEAR DECAY

Predict the products of the following nuclear reactions.

10.
$$^{236}_{92}U$$
 + $^{4}_{2}He$ \rightarrow $\frac{^{241}_{94}PU}{^{94}}$ + $^{1}_{0}n$

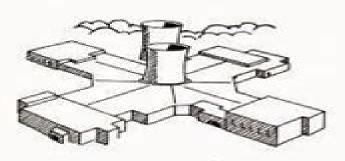
 Bombardment of aluminum-27 by alpha particles produces phosphorus-30 and one other particle. Write the nuclear equation for this reaction and identify the other particle.

12. Plutonium-239 can be produced by bombarding uranium-238 with alpha particles. How many neutrons will be produced as a by product of each reaction? Write the nuclear equation for this reaction.

 Neutron bombardment of plutonium-239 yields americium-240 and another particle. Write the nuclear equation and identify the other particle produced.

14. Alpha-particle bombardment of plutonium-239 produces a neutron and another isotope. Write the nuclear equation for this reaction and identify the isotope.

$$^{239}_{94}$$
Pu + $^{4}_{2}$ He \rightarrow $^{1}_{0}$ n + $^{242}_{96}$ Cm



Practicing Nuclear Equations Answer Key

Daniel C. Tofan

Practicing Nuclear Equations Answer Key:

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