

Balancing Equations Worksheet

- 1) ____ Na_3PO_4 + ____ KOH $\rightarrow$ ____ NaOH + ____ K_3PO_4
- 2) ____ MgF_2 + ____ Li_2CO_3 $\rightarrow$ ____ MgCO_3 + ____ LiF
- 3) ____ P_4 + ____ O_2 $\rightarrow$ ____ P_2O_3
- 4) ____ RbNO_3 + ____ BeF_2 $\rightarrow$ ____ $\text{Be}(\text{NO}_3)_2$ + ____ RbF
- 5) ____ AgNO_3 + ____ Cu $\rightarrow$ ____ $\text{Cu}(\text{NO}_3)_2$ + ____ Ag
- 6) ____ CF_4 + ____ Br_2 $\rightarrow$ ____ CBr_4 + ____ F_2
- 7) ____ HCN + ____ CuSO_4 $\rightarrow$ ____ H_2SO_4 + ____ $\text{Cu}(\text{CN})_2$
- 8) ____ GaF_3 + ____ Cs $\rightarrow$ ____ CsF + ____ Ga
- 9) ____ BaS + ____ PtF_2 $\rightarrow$ ____ BaF_2 + ____ PtS
- 10) ____ N_2 + ____ H_2 $\rightarrow$ ____ NH_3
- 11) ____ NaF + ____ Br_2 $\rightarrow$ ____ NaBr + ____ F_2
- 12) ____ $\text{Pb}(\text{OH})_2$ + ____ HCl $\rightarrow$ ____ H_2O + ____ PbCl_2
- 13) ____ AlBr_3 + ____ K_2SO_4 $\rightarrow$ ____ KBr + ____ $\text{Al}_2(\text{SO}_4)_3$
- 14) ____ CH_4 + ____ O_2 $\rightarrow$ ____ CO_2 + ____ H_2O
- 15) ____ Na_3PO_4 + ____ CaCl_2 $\rightarrow$ ____ NaCl + ____ $\text{Ca}_3(\text{PO}_4)_2$
- 16) ____ K + ____ Cl_2 $\rightarrow$ ____ KCl
- 17) ____ Al + ____ HCl $\rightarrow$ ____ H_2 + ____ AlCl_3
- 18) ____ N_2 + ____ F_2 $\rightarrow$ ____ NF_3
- 19) ____ SO_2 + ____ Li_2Se $\rightarrow$ ____ SSe_2 + ____ Li_2O
- 20) ____ NH_3 + ____ H_2SO_4 $\rightarrow$ ____ $(\text{NH}_4)_2\text{SO}_4$

Practice Balancing Equations

R Sanford

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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