

1. According to the Bronsted—Lowry definition, which chemical species can function *both* as an acid and as a base?

- Cl^-
- SO_4^{2-}
- NH_4^+
- HCO_3^-
- H_3O^+

2. In the reaction $\text{CN}^- + \text{H}_2\text{O} \rightleftharpoons \text{HCN} + \text{OH}^-$ which is an acid—base conjugate pair?

- H_2O and HCN
- H_2O and OH^-
- CN^- and H_2O
- HCN and OH^-

3. Given that HX is a stronger Bronsted acid than HY in aqueous solution, which is true of a 1 M solution of NaX ?

- It is less basic than a 1 M solution of NaY .
- It is more basic than a 1 M solution of NaY .
- It yields a neutral solution.
- It is more concentrated than a 1 M solution of NaY .

4. The pH of a 0.03 M HCl solution is

- 1.5
- 2.5
- 3.5
- 12.5

5. The pH of a 1.0×10^{-2} M $\text{Ba}(\text{OH})_2$ solution at 25°C is

- 2.7
- 3.0
- 11.0
- 11.3

6. Which is the strongest acid?

- HClO
- HClO_2
- HClO_3
- HClO_4

7. Which salt reacts with water (hydrolyzes) to produce a basic solution?

- $\text{NaC}_2\text{H}_3\text{O}_2$
- NaNO_3
- NH_4Cl
- BaSO_4

8. The *weakest* of the bases listed is

Acid	Conjugate Base	K_a (Ionization Constant of Acid)
HCl	Cl^-	100% ionized
HSO_4^-	SO_4^{2-}	1.2×10^{-2}
H_2S	HS^-	5.7×10^{-8}
HS^-	S^{2-}	1.2×10^{-13}

- Cl^-
- CN^-
- HS^-
- S^{2-}
- SO_4^{2-}

9. The oxide of which element will react with water to form the strongest acid?

	Main Groups							
Period	I	II	III	IV	V	VI	VII	O
First								
Second	X	Y	Z	P		Q	S	U
Third	W					R	T	M

- W
- M
- P
- R
- Z

10. Which statement is a logical inference from the fact that a 0.10 M solution of potassium acetate $\text{KC}_2\text{H}_3\text{O}_2$ is less alkaline than a 0.10 M solution of potassium cyanide, KCN ?

- Hydrocyanic acid is a weaker acid than acetic acid.
- Hydrocyanic acid is less soluble in water than acetic acid.
- Cyanides are less soluble than acetates.
- Acetic acid is a weaker acid than hydrocyanic acid.

11. In the titration of 50.0 mL of 0.100 M benzoic acid (a monoprotic acid) with 50.0 mL of 0.100 M NaOH , the properties of the solution at the equivalence point will correspond exactly to the properties of

- a 0.100 M sodium solution.
- a 0.0500 M sodium hydroxide solution.
- a 0.0500 M benzoic acid solution.
- a 0.0500 M sodium benzoate solution.

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in a variety of content areas and at different educational levels This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor s role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project

Organic Chemistry Education Research into Practice Jay Wackerly, Sarah Zingales, Michael Wentzel, Gautam Bhattacharyya, Brett McCollum, 2025-03-25 This Research Topic has three main goals 1 provide a platform for instructors of organic chemistry to showcase evidence based methods and educational theories they have utilized in their classrooms 2 build new and strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice There are places in the literature dedicated for chemical education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes

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Quality Assurance in the Pathology Laboratory Maciej J. Bogusz, 2011-02-22 Quality refers to the amount of the unpriced attributes contained in each unit of the priced attribute Leffler 1982 Quality is neither mind nor matter but a third entity independent of the two even though Quality cannot be defined you know what it is Pirsig 2000 The continuous formulation of good practices and procedures across fields reflects t Chemists' Guide to Effective Teaching Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry Preparing for Your ACS Examination in General Chemistry - the Official Guide LUCY T. EUBANKS, 2018-10-15 This guide is separated into first term and second term general chemistry material Each section contains 8 chapters of material that also aligns to most general chemistry textbooks for a seamless addition to study materials for students Each chapter is designed with an introductory section of the material including common representations and where to find this material in a textbook The second section provides worked examples of typical multiple choice questions including how the correct answer is determined as well as how the incorrect answers were determined Also included for each study problem is a listing of the corresponding practice questions that use that concept The final section is a series of practice problems to test the concepts collectively The key is provided on a separate page for all study and practice problems **Perspectives on Scientific Argumentation** Myint Swe Khine, 2011-09-30 Argumentation arriving at conclusions on a topic through a process of logical reasoning that includes debate and persuasion has in recent years emerged as a central topic of discussion among science educators and researchers There is now a firm

and general belief that fostering argumentation in learning activities can develop students critical thinking and reasoning skills and that dialogic and collaborative inquiries are key precursors to an engagement in scientific argumentation It is also reckoned that argumentation helps students assimilate knowledge and generate complex meaning The consensus among educators is that involving students in scientific argumentation must play a critical role in the education process itself Recent analysis of research trends in science education indicates that argumentation is now the most prevalent research topic in the literature This book attempts to consolidate contemporary thinking and research on the role of scientific argumentation in education Perspectives on Scientific Argumentation brings together prominent scholars in the field to share the sum of their knowledge about the place of scientific argumentation in teaching and learning Chapters explore scientific argumentation as a means of addressing and solving problems in conceptual change reasoning knowledge building and the promotion of scientific literacy Others interrogate topics such as the importance of language discursive practice social interactions and culture in the classroom The material in this book which features intervention studies discourse analyses classroom based experiments anthropological observations and design based research will inform theoretical frameworks and changing pedagogical practices as well as encourage new avenues of research Bulletin ,1962 *Broadening Participation in STEM* Zayika Wilson-Kennedy,Goldie S. Byrd,Eugene Kennedy, Henry T. Frierson,2019-02-28 This book reports on high impact educational practices and programs that have been demonstrated to be effective at broadening the participation of underrepresented groups in the STEM disciplines **Science** ,1969 **Catalogue** University of California, Santa Cruz, Industrial & Engineering Chemistry ,1915

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