PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2   **TIME: 1 HOUR**

VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).

1.   Write a scientific report which must include the following:
1.1   Aim   (1)
1.2   Apparatus and method   (5)

2.   **Observations and Recording of results** (Copy the table below onto your book
        or sheet of paper and transfer the results you recorded when conducting the
        experiment). Calculate   and pV for each set of results and record your answers
        in the table.

Volume (V) cm³	Pressure (p) kPa		pV (°)

                                                                                                                 (6)

3.   **Analysis of the results:**
3.1   Identify the following variables for the experiment:
        • Independent
        • Dependent
        • Controlled   (3)
3.2   Use the graph paper provided by your teacher to draw graphs of V versus p and
        V versus .   (18)

4.   **Conclusion**
        4.1   Examine the results in your table as well as the two graphs you have
              drawn, what happens to the volume of the gas in the tube as the pressure
              is decreased.   (1)
        4.2   Write down a conclusion for this experiment.   (3)

5   **Questions and discussion:**

        5.1   Explain why you have to wait for a few minutes before you can take
              readings.   (2)
        5.2   Explain the relationship between pressure and volume in terms of the
              kinetic theory of gases.   (4)

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