

# BALANCING CHEMICAL EQUATIONS

Name Grove

Rewrite and balance the equations below.

- $\text{N}_2 + \text{H}_2 \rightarrow \text{NH}_3$   $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
- $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$   $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
- $\text{NaCl} + \text{F}_2 \rightarrow \text{NaF} + \text{Cl}_2$   $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
- $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$   $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- $\text{AgNO}_3 + \text{MgCl}_2 \rightarrow \text{AgCl} + \text{Mg}(\text{NO}_3)_2$   $2\text{AgNO}_3 + \text{MgCl}_2 \rightarrow 2\text{AgCl} + \text{Mg}(\text{NO}_3)_2$
- $\text{AlBr}_3 + \text{K}_2\text{SO}_4 \rightarrow \text{KBr} + \text{Al}_2(\text{SO}_4)_3$   $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + \text{Al}_2(\text{SO}_4)_3$
- $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
- $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$
- $\text{C}_8\text{H}_{18} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   $\text{C}_8\text{H}_{18} + \frac{25}{2}\text{O}_2 \rightarrow 8\text{CO}_2 + 9\text{H}_2\text{O}$
- $\text{FeCl}_3 + \text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + \text{NaCl}$   $\text{FeCl}_3 + 3\text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + 3\text{NaCl}$
- $\text{P} + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$   $4\text{P} + 5\text{O}_2 \rightarrow 2\text{P}_2\text{O}_5$
- $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$   $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$
- $\text{Ag}_2\text{O} \rightarrow \text{Ag} + \text{O}_2$   $2\text{Ag}_2\text{O} \rightarrow 4\text{Ag} + \text{O}_2$
- $\text{S}_8 + \text{O}_2 \rightarrow \text{SO}_3$   $\text{S}_8 + 12\text{O}_2 \rightarrow 8\text{SO}_3$
- $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$   $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$
- $\text{K} + \text{MgBr}_2 \rightarrow \text{KBr} + \text{Mg}$   $2\text{K} + \text{MgBr}_2 \rightarrow 2\text{KBr} + \text{Mg}$
- $\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$   $2\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$

# Physical Science If8767 Answers Balancing Chemical

**David Kirk**



## **Physical Science If8767 Answers Balancing Chemical:**

## **Physical Science If8767 Answers Balancing Chemical** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Physical Science If8767 Answers Balancing Chemical**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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