

A chemical reaction is a transformation of a chemical substance in which the substance's chemical identity is fundamentally altered.

A coefficient represents the number of atoms or molecules present in a chemical reaction. A coefficient is placed directly in front of a chemical equation or above each atom.

A subscript is present after the element symbol to indicate the number of atoms of a given element.

Why is it important to balance chemical equations?

Chemical reactions are balanced to ensure that the number of atoms of each element is the same on both sides of the equation. This is because matter is neither created nor destroyed in a chemical reaction.

What does the law of conservation of mass state when balancing equations?

The law of conservation of mass states that the total mass of the reactants is equal to the total mass of the products in a chemical reaction.

### Week 8, Day 2

The number of protons in an atom cannot change (it is fixed) but a different number of neutrons can change.

The most stable electron configuration for atoms of each element is a complete set of valence electrons (the highest energy level) and a full set of protons. Atoms with stable electron configurations are called noble gases.

How does a positive ion form from a neutral atom? What is another name for a positive ion?

When an atom loses one or more electrons, it has more protons than it does electrons, so it is no longer neutral. This atom has a positive charge and is called a positive ion or cation.

How does a negative ion form from a neutral atom? What is another name for a negative ion?

When an atom gains one or more electrons, it has more electrons than it does protons, so it is no longer neutral. This atom has a negative charge and is called a negative ion or anion.



Atoms in Group 1A have 1 electron in their outer shell and Group 2A atoms have 2 electrons in their outer shell. To become more stable (like the noble gases), the outermost electrons are easily given up. These elements are called Group 1A and Group 2A elements. They have one or two valence electrons and easily lose them to form the most stable electron configuration of the noble gas next to them.



Atoms are formed by taking the first part of the element name and adding the suffix -ide. For example, sodium chloride (NaCl) is called a sodium chloride.



When positive and negative ions meet, the positive ions give up the electrons, becoming a neutral positively charged ion, and the negative ions become the opposite, becoming a neutral negatively charged ion.

The positively charged cation and the negatively charged anion are attracted to each other. The bond formed between the positively charged cation and the negatively charged anion is called an ionic bond.

# Physical Science Exploring Space Notetaking Guide Answer

**K Morrison**



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