



Name: \_\_\_\_\_ Date: \_\_\_\_\_ Class Period: \_\_\_\_\_

## Strength of Acids & Bases

**Directions:** The pH values of several solutions are shown below. Note: strong acids have a pH of 3 or less, and strong bases have a pH of 11 or greater.

- 1) Color the pH scale the appropriate colors to reflect the colors of litmus paper that reflect the strength of pH of solutions.
- 2) Write the name of the item on the correct location on the pH scale.
- 3) Fill in the items in the table to determine if they are a strong or weak acid or base.

### pH Scale

0	1	2	3	4	5	6	7	8	9	10	11	12	13	14
---	---	---	---	---	---	---	---	---	---	----	----	----	----	----

Strong Acids	Weak Acids	Strong Bases	Weak Bases

### pH Values:

Milk – 6.0

Soda – 3.0

Laundry detergent – 10.5

Industrial Strength Soap – 12.5

Distilled Water – 7.0

Toothpaste – 9.0

Oven cleaner – 12.5

Sodium hydroxide – 14.0

Lemon juice – 2.4

Black coffee – 5.0

Baking soda – 8.4

Hydrochloric acid – 1.0

Vinegar – 2.5

Tomato juice – 4.0

Beach – 13.0

# Physical Science 8 4 Strength Of Acids And Bases

**Tao Wei**



## **Physical Science 8 4 Strength Of Acids And Bases:**

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Macomber, 1996-04-26 This is the first of a two volume set designed for a course focused on the fundamentals of organic chemistry for pre meds and chemistry bioscience students It covers the major aspects of molecular structure followed by an introduction to the techniques of physical and organic chemistry **Chemical News and Journal of Physical Science**

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