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CLASSIFYING CHEMICAL REACTIONS

Classify the following chemical reactions as one of the following: synthesis, decomposition, single-replacement, double-replacement, or combustion.

$2H_0 + O_0 \rightarrow 2H_0O$	Caco, Cao + Co,	
3K + 3H ₂ O → H ₄ + 3KOH	CHOO, • 2HCI CHOL • HLCO.	
290010 ₉ 28001 + 30 ₉	Ca + 2HCl CaCl ₂ + H ₂	
280 _e + 0 _e 280 _e	PERNOJ, - SHCI - PECL - SHEED,	
$Ov + 2AgNO_y \rightarrow 2Ag + OutyOy$	2Na • Cl _e 2NeCli	
2H ₂ O 2H ₂ - O ₂	CH ₆ + 20 ₅ → CO ₃ + 24 ₅ O	
$PbONO_{\mathcal{O}_{S}} = 290 \rightarrow PbI_{b} = 29000_{b}$	H,CO, CO, - H,O	
2Fe + O₂ → 2FeO	Zn + 2HCt H _b + ZnCt _b	
HCI + NaOH → NaOI + HOH	25% + 254,0 25%0H + H ₂	

Pre Lab Answers To Classifying Chemical Reactions

AN Whitehead

Pre Lab Answers To Classifying Chemical Reactions:

Reviewing Pre Lab Answers To Classifying Chemical Reactions: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Pre Lab Answers To Classifying Chemical Reactions**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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