Balancing Chemical Reactions - Practice

Balance the following reactions and identify what type of reactions they are.

Type of Reaction? Synthesis



Type of Reaction? Double Replacement

$$30 1 P_4 + 3 O_2 \rightarrow 2 P_2O_3$$

Type of Reaction? Synthesis

$$43$$
 2 RbNO₃ + 1 BeF₂ \rightarrow 1 Be(NO₃)₂ + 2 RbF

Type of Reaction? Double Replacement

s)
$$2$$
 AgNO₃ + 1 Cu \rightarrow 1 Cu(NO₃)₂ + 2 Ag

Type of Reaction? Single Replacement

$$63 2 H_2O_2 \Rightarrow 2 H_2O + 1 O_2$$

Type of Reaction? Decomposition

7)
$$2$$
 HCN + 1 Cu(CN)₂

Type of Reaction? Double Replacement

$$m = 1$$
 GaF₁ + 3 Cs + 3 CsF + 1 Ga

Type of Reaction? Single Replacement

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University of Florida

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