Stoichiometry Worksheet and Key

2 KClO₃ → 2 KCl + 3 O₂

How many moles of O₂ will be formed from 1.65 moles of KClO₃?

 $\begin{array}{c|cccc}
1.65 & \text{mol KClO}_3 & \underline{\qquad} & \text{mol O}_2 \\
& \underline{\qquad} & \text{mol KClO}_3 & \underline{\qquad} & \underline{\qquad} & \text{mol O}_2
\end{array}$

2. How many moles of KClO₃ are needed to make 3.50 moles of KCl?

3.50mol KCl = ___ mol KClO₃

3. How many moles of KCl will be formed from 2.73 moles of KClO₃?

4 Fe + 3 O₂ → 2 Fe₂O₃

4. How many moles of Fe₂O₃ are produced when 0.275 moles of Fe is reacted?

0.275 mol Fe = ___ mol Fe₂O₃

5. How many moles of Fe₂O₃ are produced when 31.0 moles of O₂ is reacted?

6. How many moles of O₂ are needed to react with 8.9 moles of Fe?

 $2 H_2O \rightarrow 2 H_2 + O_2$

7. How many moles of O₂ are produced when 1.26 moles of H₂O is reacted?

How many moles of H₂O are needed to produce 55.7 moles of H₂?

If enough H₂O is reacted to produce 3.40 moles of H₂, then how may moles of O₂ must have been made?
 (a bit challenging, but just think about it and you can probably figure it out)

Practice Test Stoichiometry Answers

J Ma

 $\label{lem:practice} \textbf{Practice Test Stoichiometry Answers:}$

Decoding Practice Test Stoichiometry Answers: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Practice Test Stoichiometry Answers**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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