



NATIONAL
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2
(CHEMISTRY)
EXEMPLAR

MARKS: 100
TIME: 1 hour



The question paper consists of 10 pages, including 4 extra sheets.
Use permanent marker for writing and drawing.

QUESTION 3

PHYSICAL SCIENCES PAPER 2

INTERMOLECULAR FORCES

GRADE 11

NOVEMBER 2020

Examiner's use: _____ Physical Sciences P2

QUESTION 3 starts on a NEW page.

3.1 The boiling points in the table below were obtained during an investigation carried out to measure the boiling points of pure 1-hexanol. The table also lists the ΔH_{vap} and ΔH_{fus} for 1-hexanol in kJ mol⁻¹ and in $^{\circ}\text{C}$ respectively.

Compound	Molecular mass (M _r)	Boiling point (T _b)
A	102	162
B	102	21

- 3.1.1 Define the term boiling point. (2)
- 3.1.2 Is pure 1-hexanol composed of one (1) or two (2) and 100 (one) or 200 (two) percent? (2)
- 3.1.3 Name the type of intermolecular force that exists between molecules of pure compounds A and B and its boiling point of these molecules. (2)
- 3.1.4 What (100) of the compound (A or B) has a (100) percent lower molar enthalpy of fusion? (2)
- 3.1.5 What compound (A or B) will have the higher vapour pressure? (2)
- 3.1.6 What compound (A or B) will have the higher surface tension? (2)

Compound B (M_r 102) is shown in the table. Use a suitable reference data that will compound A and B for the following table during part of 3.1.5.

- 3.1.5 Explain why the boiling point of compound B is (100) $^{\circ}\text{C}$ during part of compound A and B for the following in the T_b and ΔH_{fus} of intermolecular forces. (2)

3.2 Add table 3.2 to add to each column of pure compounds (A) and (B) in the table below in the table below.



- 3.2.2 Explain the process in (A) or (B) about the following in the intermolecular forces. (2)

Examiner's use

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Physical Science Paper 2 Intermolecular Forces Investigation

G Orfield



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