

SOLUTION

$$w = - \int_{V_1}^{V_2} P dV \quad P = \frac{RT}{V-b} - \frac{a}{V^2}$$

$$w = - \int_{V_1}^{V_2} \frac{RT}{v-b} dV \quad \text{⊕} \quad \int_{V_1}^{V_2} \frac{a}{V^2} dV$$

$$= - RT \ln \frac{V_2-b}{V_1-b} + a \left(\frac{1}{V_1} - \frac{1}{V_2} \right)$$

✓1.9 Are the following expressions exact differentials?

(a) $xy^2 dx - x^2 y dy$

(b) $\frac{dx}{y} - \frac{x}{y^2} dy$

SOLUTION

(a) Taking the cross derivatives

$$\frac{\partial xy^2}{\partial y} = 2xy$$

$$\frac{\partial (-x^2 y)}{\partial x} = -2xy$$

Since the cross derivatives are unequal, the expression is not an exact differential.

(b) $\frac{\partial (1/y)}{\partial y} = -\frac{1}{y^2}$

$$\frac{\partial (-x/y^2)}{\partial x} = -\frac{1}{y^2}$$

Since the cross derivatives are equal, the expression is an exact differential.

- 1.10 A mole of liquid water is vaporized at 100 °C and 1.013 bar. The heat of vaporization is 40.69 kJ mol⁻¹. What are the values of (a) w_{rev} , (b) q , (c) ΔU , and (d) ΔH ?

SOLUTION

(a) Assuming that water vapor is a perfect gas and that the volume of liquid water

Physical Chemistry Robert Alberty Solution Manual

M Carnoy



Physical Chemistry Robert Alberty Solution Manual:

This book delves into Physical Chemistry Robert Alberty Solution Manual. Physical Chemistry Robert Alberty Solution Manual is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Physical Chemistry Robert Alberty Solution Manual, encompassing both the fundamentals and more intricate discussions.

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- Chapter 1: Introduction to Physical Chemistry Robert Alberty Solution Manual
- Chapter 2: Essential Elements of Physical Chemistry Robert Alberty Solution Manual
- Chapter 3: Physical Chemistry Robert Alberty Solution Manual in Everyday Life
- Chapter 4: Physical Chemistry Robert Alberty Solution Manual in Specific Contexts
- Chapter 5: Conclusion

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Table of Contents Physical Chemistry Robert Alberty Solution Manual

1. Understanding the eBook Physical Chemistry Robert Alberty Solution Manual
 - The Rise of Digital Reading Physical Chemistry Robert Alberty Solution Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Chemistry Robert Alberty Solution Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Chemistry Robert Alberty Solution Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Chemistry Robert Alberty Solution Manual
 - Personalized Recommendations
 - Physical Chemistry Robert Alberty Solution Manual User Reviews and Ratings
 - Physical Chemistry Robert Alberty Solution Manual and Bestseller Lists
5. Accessing Physical Chemistry Robert Alberty Solution Manual Free and Paid eBooks
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 - Physical Chemistry Robert Alberty Solution Manual eBook Subscription Services
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 - ePub, PDF, MOBI, and More
 - Physical Chemistry Robert Alberty Solution Manual Compatibility with Devices
 - Physical Chemistry Robert Alberty Solution Manual Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Physical Chemistry Robert Alberty Solution Manual
 - Highlighting and Note-Taking Physical Chemistry Robert Alberty Solution Manual
 - Interactive Elements Physical Chemistry Robert Alberty Solution Manual

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 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Chemistry Robert Alberty Solution Manual
9. Balancing eBooks and Physical Books Physical Chemistry Robert Alberty Solution Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Chemistry Robert Alberty Solution Manual
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Chemistry Robert Alberty Solution Manual
 - Setting Reading Goals Physical Chemistry Robert Alberty Solution Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Chemistry Robert Alberty Solution Manual
 - Fact-Checking eBook Content of Physical Chemistry Robert Alberty Solution Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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