

-

- Ⓐ Which one of the following decreases in a group from top to bottom?

51. Metallic nature

D) Electronegativity

-
- Diagram illustrating the molecular orbital configuration for the N_2^+ ion. The diagram shows the combination of atomic orbitals (1s, 2s, 2p) from two nitrogen atoms to form molecular orbitals. The resulting molecular orbital configuration is $1s^2 2s^2 2p^2$, with a total of 10 valence electrons.

-

3 - 2 = 6 M

10. Pose any two questions to understand the difference between valency and valency electrons.

11. Explain why bonding angle (HOH) in water is $104^{\circ} 31'$ instead of $109^{\circ} 28'$.

Physical Science Grade 1paper 2014

ML Yell



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