T.Y.B.Sc. Semester 5 Mock Exam Question Paper

Chemistry Paper I - Physical Chemistry

Multiple Choice Questions

- Q1 Vapour pressure of a liquid solventwhen a non volatile solute is 1M dissolve in it.
 - a) Increase
 - b) Decrease
 - c) no change
 - d) Expand
- O2 Raoult's law is expressed by......

a)
$$\underline{P}^{\alpha} - \underline{P} = X_2$$

c)
$$P - P^0 = X_2$$

$$d) \underline{P \cdot P^{\circ}} = X$$

- O3is used in Rast method because it has a high value of K₁.
 - a) Camphor
 - b) Bi phenyl
 - c) Naphthalene
 - d) Sodium chloride
- Q4 Boiling point elevation of the solution ΔT_b is given as.......
 - a) $\Delta T_b = T T_o$
 - b) $\Delta T_b = T_o T$
 - c) $\Delta T_{t_0} = T + T_0$
 - d) $\Delta T_b = T \times T_0$
- Q5 The unit for activation of energy is _____
 - a) kJ mof⁻¹
 - b) kJ К-¹
 - c) kJ K-1 mol-1
 - d) kJ R-¹

Physical Chemistry Multiple Choice Questions

Alyn G. McFarland, Nora Henry

Physical Chemistry Multiple Choice Questions:

Multiple Choice Questions in Physical Chemistry for Undergraduate Students Dr Kusum Baser, 2022-08-25 Objective questions are very much useful for practice of particular topic Multiple choice questions are important to practice any topic to solve objective questions very deep study and knowledge of specific topic is needed. This book helps the students to do practice in such a way so they can do preparation for competitive exams This book covers important topics of physical chemistry for undergraduate students Physical Chemistry: A Level Multiple Choice Questions Margaret Marion Langley Oblitas, 1970-01-01 Physical Chemistry M.M.L. Oblitas, 1971 Physical Chemistry M.M.L. Oblitas, 1977 Method Approach for Introductory Physical Chemistry Calculations Brian J Hathaway, Clair Murphy, Brian Murphy, 2019-05-16 A Working Method Approach for Introductory Physical Chemistry Calculations is a concise inexpensive introduction to first year chemistry that is aimed at students who are weak in chemistry or have no chemistry on entry to university Such students usually find physical chemistry the most difficult part of the chemistry course and within this section numerical problem solving is an additional difficulty The text should also be invaluable to first year intending chemists This text provides an introduction to physical chemistry and the gas laws followed by chapters on thermodynamics chemical equilibrium electrochemistry and chemical kinetics Each section involves a brief introduction followed by a representative examination question which is broken down into a proposed working method Both short multiple choice questions and related full examination type questions are included This book will prove invaluable to students who need encouragement in a logical approach to problem solving in physical chemistry teaching them to think for themselves when faced with a problem

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Multiple Choice Questions in A-level Chemistry B. P. Bourne, 1969

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