#### EARTH AND LIFE SCIENCE 2<sup>ND</sup> QUARTER G12 ABM- REVIEWER

#### LESSON 1:

#### Exogenic Processes

- It is a blanket term for those processes which operate on or close to the surface of the Earth.
- It includes geological phenomena and processes that originate externally to the Earth's surface.

#### Types of Exogenic Processes

#### Weathering

- It is the breaking down of rocks into fragments or into smaller size.
- Every rock had undergone weathering, whether it is disintegration (physical weathering) or decomposition (chemical weathering).

#### Physical Weathering

 Refers to the breakdown of rocks into pieces without any change in its composition.

#### Types of Physical Weathering

#### Unloading

- It is the uplifted intrusive igneous rocks.
- It means rocks from beneath the surface go above the surface.
- Weathering are caused by variety of factors such as tectonism and volcanism.

#### Thermal Expansion & Contractions

- It refers to the effect of temperature to rocks.
- When exposed to high temperature, rocks expand; when exposed to low temperature, rocks contract

#### Freeze- Thow Weathering

- It is most common in the North pole and South.
- It occurs when a fractured rock absorbs water through its crack then got frozen.
- As mention in Thermal Expansion and contraction, the colder the temperature, the molecule in it expand.

#### Hydration

- It refers to the amount of water being absorbed by the rock.
- > When hydrated too much or plenty of water is present the rock softens.
- When it is dehydrated too less or lack of waterthe rock hardens.
- > The repeated process of this causes weathering

#### Salt Crystal Growth

- > It usually happens in the sea or the seaside.
- It happens when salt water comes in contact with rocks.
- The wind can also take part in the salt crystal growth.

#### Chemical Weathering

Refers to the changes in the composition of rock.

#### Types of Chemical Weathering

#### Oxidation

- O\* in H\*O comes in contact with certain elements in minerals of rocks, and chemical reaction can occur.
- ➤ E.g. Fe and Al Fe<sup>2</sup> O<sup>2</sup> and Al<sup>2</sup> O<sup>2</sup>

#### Hydrolysis

- H°O molecules alone or CO° in H°O react with chemical composition/components of rock minerals to create new components or compounds.
- E.g. silicate minerals (containing silica and O\*) in igneous rock silicate minerals – clay minerals = KalSi\* O\* + H\*O – HalSi
- > Reactions that produces weak acid.

#### Solution

- > Rock forming mineral dissolve in water
- E.g. halite mineral containing rock salt (NaCl) susceptible to solution in H\*O

#### Carbonation

- Involves CO\* and H\*O molecules reacting with, and thereby decomposing rock material.
- ➤ E.g. carbonate rocks (CO²) like limestone (composed of CaCO²) (O² + H²O - H² CO² (carbonic acid) CaCO (limestone) + H² CO² (carbonic acid) + Ca HCO²) calcium bicarbonate

#### Mass Wasting

- > also called mass movement
- downslope transport of surface materials in direct response to gravity.
- heavier objects have greater downward materials in direct response gravity.
- the force of gravity encourages rock sediments and soil to move downhill or sloping surfaces.

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