OXIDATION-RERDUCTION BALANCING

Balance the following redox reactions in acidic solutions.

$$0 \text{ Sn}^{2*} + 10^{-} \rightarrow \text{Sn}^{4*} + 1^{-}$$

Balance the following redox reactions in basic solutions.

(M)
$$MnO_4$$
: $+C_2O_4$ ² $\rightarrow MnO_2$ $+CO_2$

$$(x)$$
 Te + NO, \rightarrow TeO, 2 + N,O,

Oxidation Number Balancing Equations Worksheet

O. Jerry Parker, Gary L. Breneman

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