Answer key-H.W(2)-Oxidation-reduction reaction

1-Multiple choice questions:

1-Oxidation involves

a-gaining electrons; gaining oxygen b-losing electrons; losing oxygen c-gaining electrons; losing oxygen d-losing electrons; gaining oxygen

2- Reduction involves

a-gaining electrons; gaining oxygen b-losing electrons; losing oxygen c-gaining electrons, losing oxygen d-losing electrons; gaining oxygen

3-In the reaction

2Ca(s) + O2(g) --> 2CaO(s), calcium is _____ a-Reduced b- Synthesized c-Oxidized

4- In the reaction Zn + H₂O --> ZnO₂ + H₂ which element, is oxidized?

a-Zinc b-Hydrogen c-Oxygen

5- Which substance is reduced in the following reaction?

NaOH + Li --> LiOH + Na

a-U <u>b-Na</u> c-O d-H

6-Which of the following is the balanced half -reaction for the oxidation of Cu to Cu*2?

a-Cu → Cu⁺² b-Cu+2e- → Cu⁺² c-Cu → Cu⁺² +2e-

7-Another name for a "oxidation-reduction" reaction is

a-chemical reaction b-neutralization reaction c-redox reaction

Remember

Redox (Oxidation and Reduction)

Gain oxygen †	Loss oxygen
Gain oxidation Number †	Loss exidation Number
Loss hydrogen;	Gain hydrogen

Oxidation And Reduction Packet Answer Key

James Woodward Kent

Oxidation And Reduction Packet Answer Key:

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