

**Section 1.1** *continued*

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	1
_____	2
_____	3
_____	4
_____	5

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 16. It exists in the periodic table.
- _____ 17. It has no line periodic table.
- _____ 18. Group 1 elements.
- _____ 19. Elements that are strong oxidizing agents.
- _____ 20. Group 8 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 21. There are two representative elements of elements.
- _____ 22. Most elements have lines in the periodic table for metals.
- _____ 23. Group 1 elements except for hydrogen are metals in the solid state.
- _____ 24. Group 2 elements are the alkaline earth metals.
- _____ 25. Group 1 elements are highly reactive elements known as alkalis.
- _____ 26. Group 8 elements are very reactive elements known as noble gases.
- _____ 27. Elements have properties of both metals and non-metals.

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B Lingard



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