

- 10. The reaction is exothermic because the products have a lower energy than the reactants.
- 11. The activation energy is the energy difference between the reactants and the products.
- 12. The energy of the reactants is higher than the energy of the products.
- 13. The energy of the products is lower than the energy of the reactants.



Figure 1



Figure 2

Figure 1 shows the potential energy diagram for an endothermic reaction. Figure 2 shows the potential energy diagram for an exothermic reaction.

Answer the questions in the margin.

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Table of Contents Sca2hp Science Paper Unit January

1. Understanding the eBook Sca2hp Science Paper Unit January
 - The Rise of Digital Reading Sca2hp Science Paper Unit January
 - Advantages of eBooks Over Traditional Books
2. Identifying Sca2hp Science Paper Unit January
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Sca2hp Science Paper Unit January
 - User-Friendly Interface
4. Exploring eBook Recommendations from Sca2hp Science Paper Unit January
 - Personalized Recommendations

- Sca2hp Science Paper Unit January User Reviews and Ratings
- Sca2hp Science Paper Unit January and Bestseller Lists
- 5. Accessing Sca2hp Science Paper Unit January Free and Paid eBooks
 - Sca2hp Science Paper Unit January Public Domain eBooks
 - Sca2hp Science Paper Unit January eBook Subscription Services
 - Sca2hp Science Paper Unit January Budget-Friendly Options
- 6. Navigating Sca2hp Science Paper Unit January eBook Formats
 - ePub, PDF, MOBI, and More
 - Sca2hp Science Paper Unit January Compatibility with Devices
 - Sca2hp Science Paper Unit January Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Sca2hp Science Paper Unit January
 - Highlighting and Note-Taking Sca2hp Science Paper Unit January
 - Interactive Elements Sca2hp Science Paper Unit January
- 8. Staying Engaged with Sca2hp Science Paper Unit January
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Sca2hp Science Paper Unit January
- 9. Balancing eBooks and Physical Books Sca2hp Science Paper Unit January
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Sca2hp Science Paper Unit January
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Sca2hp Science Paper Unit January
 - Setting Reading Goals Sca2hp Science Paper Unit January
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Sca2hp Science Paper Unit January
 - Fact-Checking eBook Content of Sca2hp Science Paper Unit January

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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