# Polyatomic Ions Chart

Formula	Name	Formula	Name
NH4°	Ammonium	CrO <sub>4</sub> -2	Chromate
NH <sub>3</sub>	Ammonia	Cr2O7-2	Dichromate
C2H3O2	Acetate	MnO <sub>4</sub>	Permanganate
CH₃COO-	Acetate	MnO <sub>4</sub> -2	Manganate
CN-	Cyanide	NO <sub>2</sub>	Nitrite
CO3-2	Carbonate	NO₃ <sup>-</sup>	Nitrate
HCO3	Bicarbonate	OH-	Hydroxide
C2O4-2	Oxalate	PO <sub>4</sub> -3	Phosphate
CIO	Hypochlorite	SCN-	Thiocyanate
CIO2-	Chlorite	Fe(CN) <sub>6</sub> -3	Ferricyanide
CIO3	Chlorate	5O₃ <sup>-2</sup>	Sulfite
CIO4	Perchlorate	5O <sub>4</sub> -2	Sulfate
5 <sub>2</sub> O <sub>3</sub> -2	Thiosulfate	H5O <sub>4</sub> -	Hydrogen sulfate
BrO-	Hypobromite	IO3-	Iodate
AsO2-3	Arsenite	SeO <sub>4</sub> -2	Selenate
BrO <sub>3</sub>	Bromate	HSO₃⁻	Hydrogen sulfite

# **Polyatomic Ions Extensions**

**SB Merriam** 

**Polyatomic Ions Extensions:** 

Unveiling the Magic of Words: A Overview of "Polyatomic Ions Extensions"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Polyatomic Ions Extensions**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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