

Bonding and Reactions

Study Guide

10 pages including KEY

Chemical Bonding and Reactions Study Guide

This study packet will help you study for the Chemical Bonding and Reactions exam by highlighting the most important concepts in the course. This packet highlights key concepts and their correct application in the course and should be used as a guide to the material and not a replacement.

B. VOCABULARY

Atom	Electron	Electron transfer	Electron transfer
Atomic mass	Electron	Electron transfer	Electron transfer
Atomic weight	Electron	Electron transfer	Electron transfer
Atomic	Electron	Electron transfer	Electron transfer
Chemical bond	Electron	Electron transfer	Electron transfer
Chemical reaction	Electron	Electron transfer	Electron transfer

C. READING

1. Why is the bond length important?

2. What is the bond length?

3. What kind of electron flow is the bond?

4. What is a chemical bond?

5. What kind of electron flow is the bond?

6. Why don't noble gases bond with other elements?

Why don't noble gases bond with other elements?

1. Noble gases have a full outer shell of electrons.

2. Noble gases have a full outer shell of electrons.

Why don't noble gases bond with other elements?

1. Noble gases have a full outer shell of electrons.

2. Noble gases have a full outer shell of electrons.

REACTIONS

What happens in a chemical reaction?

1. A chemical reaction is a process in which one or more substances are converted into one or more different substances.

2. A chemical reaction is a process in which one or more substances are converted into one or more different substances.

3. A chemical reaction is a process in which one or more substances are converted into one or more different substances.

4. A chemical reaction is a process in which one or more substances are converted into one or more different substances.

5. A chemical reaction is a process in which one or more substances are converted into one or more different substances.

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7. A chemical reaction is a process in which one or more substances are converted into one or more different substances.

8. A chemical reaction is a process in which one or more substances are converted into one or more different substances.

9. A chemical reaction is a process in which one or more substances are converted into one or more different substances.

10. A chemical reaction is a process in which one or more substances are converted into one or more different substances.



B. VOCABULARY

1. Atom: A negatively charged particle that is the basic unit of matter.

2. Atomic mass: The mass of an atom, measured in atomic mass units (amu).

3. Atomic weight: The average atomic mass of an element, measured in amu.

4. Chemical bond: A force that holds atoms together in a molecule or compound.

5. Chemical reaction: A process in which one or more substances are converted into one or more different substances.

6. Electron: A negatively charged particle that is the basic unit of matter.

7. Electron transfer: A process in which one or more electrons are transferred from one atom to another.

Physical Science Bonding And Chemical Reactions Test

Marcel A. Müller



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