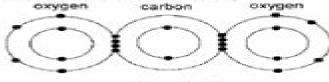


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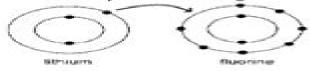
- I. MULTIPLE CHOICE INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)
- 1. It is a subatomic particle, symbol e- or
  - β-, with a negative elementary electric charge.
    - A. Neutron
    - 8. Electron
    - C. Protom
    - D. Attorni
- It is a subatomic particle, symbol p or p+, with a
  positive electric charge of +1e elementary
  charge and mass slightly less than that of a
  meutron.
  - A. Neutron
  - 8. Electron
  - C. Protom
  - D. Altern
- It is a subatomic particle, symbol n or nO, with no net electric charge and a mass slightly larger than that of a proton.
  - A. Neutron B. Electron
  - C. Proton
  - D. Atom
- 4. Is this an example of ionic bonding?



carbon dioxide

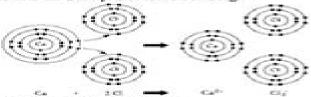
- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- C. There is no chemical bonding present.
- D. This is just an orbital model.

5. Is this an example of ionic bonding?



littreum fluoride

- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- D. This is just an orbital model.
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
  - A. Electron Configuration
  - Periodic Table of Elements
  - C. Orbital Models
  - D. Emergy Levels
- 7. Is this an example of ionic bonding?



- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.
- It is a chemical rule of thumb that reflects observation that atoms of main-group elements tend to combine in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas.
  - A. Octet Bulle
  - 8. Valence Electrons
  - C. Atomic Orbitals
  - D. Chemical Bonds
- It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
  - A. Neutron
  - B. Electron
  - C. Proton
  - D. Atom
- 10. It is a type of chemical bond that involves the electrostatic attraction between oppositely

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**L** Reisser

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- 1. Understanding the eBook Physical Science Grade 11 Final Examination 2014 Scope
  - The Rise of Digital Reading Physical Science Grade 11 Final Examination 2014 Scope
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Physical Science Grade 11 Final Examination 2014 Scope
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Grade 11 Final Examination 2014 Scope
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Physical Science Grade 11 Final Examination 2014 Scope
  - Personalized Recommendations
  - Physical Science Grade 11 Final Examination 2014 Scope User Reviews and Ratings
  - Physical Science Grade 11 Final Examination 2014 Scope and Bestseller Lists

- 5. Accessing Physical Science Grade 11 Final Examination 2014 Scope Free and Paid eBooks
  - Physical Science Grade 11 Final Examination 2014 Scope Public Domain eBooks
  - Physical Science Grade 11 Final Examination 2014 Scope eBook Subscription Services
  - Physical Science Grade 11 Final Examination 2014 Scope Budget-Friendly Options
- 6. Navigating Physical Science Grade 11 Final Examination 2014 Scope eBook Formats
  - o ePub, PDF, MOBI, and More
  - Physical Science Grade 11 Final Examination 2014 Scope Compatibility with Devices
  - Physical Science Grade 11 Final Examination 2014 Scope Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Physical Science Grade 11 Final Examination 2014 Scope
  - Highlighting and Note-Taking Physical Science Grade 11 Final Examination 2014 Scope
  - Interactive Elements Physical Science Grade 11 Final Examination 2014 Scope
- 8. Staying Engaged with Physical Science Grade 11 Final Examination 2014 Scope
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science Grade 11 Final Examination 2014 Scope
- 9. Balancing eBooks and Physical Books Physical Science Grade 11 Final Examination 2014 Scope
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Grade 11 Final Examination 2014 Scope
- 10. Overcoming Reading Challenges
  - o Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Grade 11 Final Examination 2014 Scope
  - Setting Reading Goals Physical Science Grade 11 Final Examination 2014 Scope
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Grade 11 Final Examination 2014 Scope
  - Fact-Checking eBook Content of Physical Science Grade 11 Final Examination 2014 Scope
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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