

- 1109.** When a charged particle in motion enters in a uniform magnetic field perpendicularly then its  
 (1) Speed changes  
 (2) Velocity changes  
 (3) K.E. changes  
 (4) Acceleration does not change
- 1110.** The frequency of seconds pendulum is  
 (1) 0.5 Hz (2) 1.0 Hz  
 (3) 2.0 Hz (4) 1.5 Hz
- 1111.** The structure of solids is investigated by using  
 (1) Cosmic rays (2) X-rays  
 (3) Gamma rays (4) Infrared rays
- 1112.** Two bodies with kinetic energy in the ratio of 4 : 1 are moving with equal linear momentum. The ratio of their masses is  
 (1) 1 : 2 (2) 1 : 1 (3) 4 : 1 (4) 1 : 4
- 1113.** A ray of light is incident on the surface of separation of a medium with the velocity of light in air at an angle  $45^\circ$  and is refracted in the medium at an angle  $30^\circ$ . What will be the velocity of light in the medium  
 (1)  $1.96 \times 10^8$  m/s (2)  $2.12 \times 10^8$  m/s  
 (3)  $3.18 \times 10^8$  m/s (4)  $3.33 \times 10^8$  m/s
- 1114.** Analyse the given statements and choose the correct option.  
**Statement-I :** When current is represented by a straight line, the magnetic field will be circular.  
**Statement-II :** According to Fleming's left hand rule, the direction of the force is parallel to the magnetic field.  
 (1) Both statement-I and statement-II are correct and statement-II is the correct explanation of statement-I.  
 (2) Both statement-I and statement-II are true but statement-II is not the correct explanation of statement-I.  
 (3) Statement-I is true but statement-II is false.  
 (4) Statement-I is false but statement-II is true.
- 1115.** Which one of the following is the smallest in size?  
 (1)  $\text{N}^{3-}$  (2)  $\text{O}^{2-}$  (3)  $\text{F}^-$  (4)  $\text{Na}^+$
- 1116.** Which chemical substance is added to LPG to help in detection of its leakage?  
 (1) Isobutane (2) Ethanethiol  
 (3) Propane (4) Hydrogen sulphide

- 1117.** Which of the following salt does not contains the water of crystallization?  
 (1) Blue Vitriol (2) Baking soda  
 (3) Washing soda (4) Gypsum
- 1118.** The ion of an element has 3 positive charge, 27 mass-number and 14 neutrons. What is the number of electrons in this ion?  
 (1) 13 (2) 14 (3) 10 (4) 16
- 1119.** Which one of the following is the natural fruit ripening hormone?  
 (1) Ethane (2) Ethene  
 (3) Ethyne (4) Carbide
- 1120.** Which of the following contains acidic hydrogen?  
 (1) Ethene (2) Ethane  
 (3) Ethyne (4) But-2-yne
- 1121.** Silver jewellery becomes black on prolonged exposure to air. It is due to the formation of  
 (1)  $\text{Ag}_3\text{N}$  (2)  $\text{Ag}_2\text{O}$   
 (3)  $\text{Ag}_2\text{S}$  and  $\text{Ag}_3\text{N}$  (4)  $\text{Ag}_2\text{S}$
- 1122.** What is the mass of oxygen required to react completely with 15 g of  $\text{H}_2$  gas to form water?  
 (1) 140 g (2) 115 g (3) 107.5 g (4) 120 g
- 1123.** Percentage purity of a sample of gold is 85%. How many atoms of gold are present in its 1 gram sample? (Atomic mass of gold = 197 u)  
 (1)  $2.6 \times 10^{21}$  (2)  $2.6 \times 10^{23}$   
 (3)  $3.0 \times 10^{21}$  (4)  $4.5 \times 10^{23}$
- 1124.** The volume of 1 mole of an ideal gas at  $25^\circ\text{C}$  temperature and 1 bar pressure is:-  
 (1) 22.4 L (2) 22.7 L  
 (3) 24.8 L (4) 24.4 L
- 1125.** Which of the following solution can be stored in aluminium container?  
 (1)  $\text{MgSO}_4(\text{aq})$  (2)  $\text{ZnSO}_4(\text{aq})$   
 (3)  $\text{CuSO}_4(\text{aq})$  (4)  $\text{FeSO}_4(\text{aq})$
- 1126.** What is the correct order of pH of aqueous solution of the following salts?  
 (1)  $\text{NaCl} = \text{Na}_2\text{CO}_3 = \text{NH}_4\text{Cl}$   
 (2)  $\text{NaCl} < \text{Na}_2\text{CO}_3 < \text{NH}_4\text{Cl}$   
 (3)  $\text{NH}_4\text{Cl} < \text{Na}_2\text{CO}_3 < \text{NaCl}$   
 (4)  $\text{NH}_4\text{Cl} < \text{NaCl} < \text{Na}_2\text{CO}_3$

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