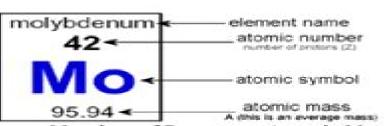
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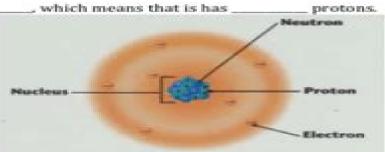
#### SPS1: Students will investigate our current understanding of the atom.

#### a. Examine the structure of the atom in terms of

- proton, electron, and neutron locations.
  - Proton has a charge of \_\_\_\_\_\_ and is located in the \_\_\_\_\_\_ of the atom.
    - Protons are also the elements number of the identity of the element, so protons are found by looking at the atomic number
  - Neutron has a charge of \_\_\_\_\_\_ and is located in the \_\_\_\_\_ of the atom.
    - Neutrons are found by subtracting the \_\_\_\_\_\_ Number from the \_\_\_\_\_\_ Mass or the BIG #-SMALL #.
  - Electron has a charge of \_\_\_\_\_ and is located in the \_\_\_\_ of the atom.
    - Electrons are equal to the number of \_\_\_\_\_\_ in a neutral atom.
  - The nucleus is made up of these 2 particles\_\_\_\_\_\_ and \_\_\_\_\_
  - The electrons are located in this region of the atom, known as the \_\_\_\_\_\_
  - The electron cloud is divided in to energy levels or energy shells, the first level will hold electrons and the second level will hold \_\_\_\_\_\_ electrons.
  - The valence shell or valence level is the \_\_\_\_\_\_ energy shell of the atom.
- atomic mass and atomic number.
  - atoms identity-it's the small number of the periodic table.
  - Atomic mass is equal to the average mass of all the isotopes of a particular element—it the big number on the periodic table.
- atoms with different numbers of neutrons (isotopes).
  - Define Isotope
  - Ex. Carbon-12 and Carbon-14
    - Carbon 12 has \_\_\_\_P \_\_\_N \_\_\_E Carbon 14 has \_\_\_P \_\_\_N \_\_\_E
- explain the relationship of the proton number to the element's identity.
  - The \_\_\_\_\_\_ number is equal to the number of \_\_\_\_\_ in the atom which is how the element is identified.
  - Ex. Nitrogen has an atom number of , which means that is has protons.



- Number of Protons = Atomic Number
- Number of Electrons = Number of Protons = Atomic Number
- Number of Neutrons = Mass Number Atomic Number



# **Physical Science Eoct Study Guide Forsyth**

**Tobias Bleicker** 

**Physical Science Eoct Study Guide Forsyth:** 

The book delves into Physical Science Eoct Study Guide Forsyth. Physical Science Eoct Study Guide Forsyth is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Physical Science Eoct Study Guide Forsyth, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
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  - Chapter 3: Physical Science Eoct Study Guide Forsyth in Everyday Life
  - Chapter 4: Physical Science Eoct Study Guide Forsyth in Specific Contexts
  - ∘ Chapter 5: Conclusion
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web an innovative and prolific master in three media he is generally considered one of the greatest visual artists in the history of art and the most important in dutch art history unlike most dutch masters of the 17th century rembrandt s works depict a wide range of style and subject matter from portraits and self portraits to landscapes

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web we know that at least two of rembrandt s pupils went into independent art careers govaert flinck and ferdinand bol their art reflects to some degree rembrandt s influence which is expected as they learned to paint in his style however with time their style developed into their own list of rembrandt s pupils and collaborators