

- a) Benzene.
- b) Octane.
- c) Pentane.
- d) Entanol.

ANSWER : Pentane.

39) When a solute is dissolved in water it shows:

- a) decrease in freezing point of water.
- b) decrease in boiling point of water.
- c) increase in vapour pressure of water.
- d) all of the above.

ANSWER : decrease in freezing point of water.

40) When sugar is added to water, what is the change observed in boiling and freezing points of water?

- a) Both boiling point and freezing point decreases.
- b) Both boiling point and freezing point increases.
- c) Boiling point increases and freezing point decreases.
- d) Boiling point decreases and freezing point increases.

ANSWER : Boiling point increases and freezing point decreases.

41) aqueous molal solution have highest freezing point.

- a) Urea
- b) Barium chloride.
- c) Potassium bromide.
- d) Aluminium sulphate.

ANSWER : Urea

42) When a thin slice of sugar beet is placed in a concentrated solution of NaCl then sugar beet will.....

- a) dissolve in solution.
- b) neither absorb nor lose water.
- c) absorb water from solution.
- d) lose water from its cell.

ANSWER : lose water from its cell.

43) The wrong relation between osmotic pressure (P), volume (V) and temperature (T) is:

- a) $P \propto n$ if T and V are constant.
- b) Pv is constant if T is constant.
- c) $P \propto V$ if T is constant.
- d) $P \propto T$ if V is constant.

ANSWER : $P \propto V$ if T is constant.

44) Two solutions C and D are separated by a semi-permeable membrane. If liquid flows from D to C then.

- a) Both have same concentration.
- b) D is less concentrated than C.
- c) D is more concentrated than C.
- d) None of these.

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Scott C. Dulebohn



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