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Tribhuvan University
Institute of Science and Technology
B.Sc. Entrance Examination
Model question

Full marks: 100

Time: 2 hrs

Attempt all questions.

Group A (Chemistry)

(25 x 1 = 25)

- Which of the following elements does not contain neutron?
 - H
 - He
 - Be
 - Li
- The oxidation number of Nitrogen is - 1 in
 - NH₃
 - NH₂OH
 - N₂
 - N₂O
- The number of unpaired electrons in oxygen is
 - 2
 - 3
 - 4
 - 6
- The unit of radioactivity is
 - Fajan
 - Bohr
 - Curie
 - Debye
- Which of the following compounds gives positive AgNO₃ test?
 - CCl₄
 - CH₃Cl
 - CHCl₃
 - NaCl
- 'At same temperature and pressure equal volume of all gases has same number of molecules' this statement belongs to
 - Boyle's law
 - Avogadro's law
 - Charles' law
 - Graham's law
- A gas diffuses 4 times as fast as O₂. The molecular weight of the gas is
 - 4..2
 - 8
 - 16
 - 2
- In a solution containing 1 mole alcohol and 4 mole water, the mole fraction of alcohol is
 - 1/4
 - 3/4
 - 4/5
 - 1/5

Sample Question Papers Of Ee3g

Jan de Boer, H. J. Mang



Sample Question Papers Of Ee3g:

Tokyo Astronomical Observatory Reprints Tōkyō Tenmondai, 1941 *American Printer and Bookmaker* , 1925

Proceedings of the International Conference on Nuclear Physics Jan de Boer, H. J. Mang, 1973

This book delves into Sample Question Papers Of Ee3g. Sample Question Papers Of Ee3g is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Sample Question Papers Of Ee3g, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Sample Question Papers Of Ee3g
 - Chapter 2: Essential Elements of Sample Question Papers Of Ee3g
 - Chapter 3: Sample Question Papers Of Ee3g in Everyday Life
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 - Chapter 5: Conclusion
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