Name: Class:_ Five Types of Chemical Reaction Worksheet Answers Balance the following reactions and indicate which of the six types of chemical reaction are being 2 NaBr + 1 Ca(OH)₂ → 1 CaBr₂ + 2 NaOH Type of reaction: double displacement að. 2 NH₁+ 1 H₂SO₄ → 1 (NH₄)₂SO₄ **b**) Type of reaction: synthesis 4 C₅H₉O + 27 O₂ → 20 CO₂ + 18 H₂O Type of reaction: combustion CB. di) 3 Pb + 2 H₂PO₄ → 3 H₂ + 1 Pb₁(PO₄)₂ Type of reaction: single displacement Type of reaction: double displacement et. 1 Li₂N + 3 NH₄NO₃ → 3 LiNO₃ + 1 (NH₄)₃N Section 3: Predicting the products of chemical reactions 1) 2 Ag + 1 CuSO₄ → 1 Ag₂SO₄ + 1 Cu Type: Single Displacement 23 2 Nai + 1 CaCl₂ → 2 NaCl + 1 Cal₂ Type: Double Displacement 30 1 O2 + 1 H2 → 2 H2O Type: Synthesis 400 2 AgNO₂ + 1 BaSO₄ → 1 Ag₂SO₄ + 1 Ba(NO₂)₂ Type: Double Displacement 51 2 HCN + 1 CuSO₄ → 1 H₂SO₄ + 1 Cu(CN)₂ Type: Double Displacement 6) 1 H₂O + 1 AgI → 1 HI + 1 AgOH Type: Double Displacement 73 4 LiBr + 1 Co(SO₁)₂ → 2 Li₂SO₁ + 1 CoBr₄ Type: Double Displacement

Type: Single Displacement

Type: Single Displacement

Type: Single Displacement

Type: Double Displacement

Type: Single Displacement

Type: Double Displacement

Type: Decomposition

Type: Synthesis

Type: Decomposition

Type: Combustion

Type: Combustion

8):

9)

1.0%

1.13

12)

13)

14)

151

1.6)

171

18)

1 LiNO₁ + 1 Ag → 1 AgNO₁ + 1 Li

1 N2 + 2 O2 → 2 NO2

1 H₂CO₃ → 1 CO₂ + 1 H₂O

1 AlCl₃ + 3 Cs → 3 CsCl + 1 Al

1 C₀H₀ + 6 O₂ → 4 CO₂ + 4 H₂O

1 BaS + 1 PtCl₂ → 1 BaCl₂ + 1 PtS

2 Na₂O → 4 Na + 1 O₂

1 Al(NO₃)₃ + 1 Ga → 1 Ga(NO₃)₃ + 1 Al

2 KCl + 1 Mg(OH)₂ → 2 KOH + 1 MgCl₂

1 Zn + 1 Au(NO₂)₂ → 1 Zn(NO₂)₂ + 1 Au

1 CH₃COOH + 2 O₂ → 2 CO₂ + 2 H₂O

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