

EXPERIMENT 9 **CHEMICAL REACTIONS: STOICHIOMETRY** **LABORATORY REPORT**

Your Name _____ TA's Name _____ Lab Section 1411. _____ Date _____

You must read pages 69 and 70 to complete the following calculations.

	Trial #1	Trial #2	Trial #3
Vol. 0.100 M K_2CrO_4 solution	<u>12.00 mL</u>	<u>12.00 mL</u>	<u>12.00 mL</u>
mmole K_2CrO_4	<u>1.20</u>		
Vol. 0.100 M $Pb(NO_3)_2$ solution	<u>10.00 mL</u>	<u>12.00 mL</u>	<u>14.00 mL</u>
mmole $Pb(NO_3)_2$	<u>1.0</u>		
mass product + paper	<u>0.579 g</u>	<u>0.654 g</u>	<u>0.639 g</u>
mass dry filter paper	<u>0.256 g</u>	<u>0.266 g</u>	<u>0.251 g</u>
mass $PbCrO_4$ collected precip.	<u>.323</u>		
mmole $PbCrO_4$	<u>1.0 mmole</u>	<u>1.2</u>	

CALCULATIONS AND QUESTIONS

(USE THE SPACE ON THE BACK OF THIS SHEET IF NECESSARY)

1. In each trial determine which reagent was the limiting reagent? Give a reason, determine the results of your experiment, which proves your conclusion.



2. Calculate the number of millimoles of K_2CrO_4 in 10.55 mL of a 0.120 M solution. Report answer with 3 significant figures (SF).

Reaction Stoichiometry Lab Answers

Colleen F. Craig, Kim N. Gunnerson



Reaction Stoichiometry Lab Answers:

Lab Manual for General, Organic, and Biochemistry Denise Guinn, Rebecca Brewer, 2009-08-21 Teaching all of the necessary concepts within the constraints of a one term chemistry course can be challenging Authors Denise Guinn and Rebecca Brewer have drawn on their 14 years of experience with the one term course to write a textbook that incorporates biochemistry and organic chemistry throughout each chapter emphasizes cases related to allied health and provides students with the practical quantitative skills they will need in their professional lives *Essentials of General Organic and Biochemistry* captures student interest from day one with a focus on attention getting applications relevant to health care professionals and as much pertinent chemistry as is reasonably possible in a one term course Students value their experience with chemistry getting a true sense of just how relevant it is to their chosen profession To browse a sample chapter view sample ChemCasts and more visit www.whfreeman.com/gob

Optimizing STEM Education With Advanced ICTs and Simulations Levin, Ilya, Tsybulsky, Dina, 2017-06-05 The role of technology in educational settings has become increasingly prominent in recent years When utilized effectively these tools provide a higher quality of learning for students *Optimizing STEM Education With Advanced ICTs and Simulations* is an innovative reference source for the latest scholarly research on the integration of digital tools for enhanced STEM based learning environments Highlighting a range of pivotal topics such as mobile games virtual labs and participatory simulations this publication is ideally designed for educators professionals academics and students seeking material on emerging educational technologies

Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

Research Based Undergraduate Science Teaching Dennis W. Sunal, Cynthia S Sunal, Emmett L. Wright, Cheryl L. Mason, Dean Zollman, 2014-07-01 *Research in Science Education RISE Volume 6 Research Based Undergraduate Science Teaching* examines research theory and practice concerning issues of teaching science with undergraduates This RISE volume addresses higher education faculty and all who teach entry level science The focus is on helping undergraduates develop a basic science literacy leading to scientific expertise RISE Volume 6 focuses on research based reforms leading to best practices in teaching undergraduates in science and engineering The goal of this volume is to provide a research foundation for the professional development of faculty teaching undergraduate science Such science instruction should have short and longterm impacts on student outcomes The goal was carried out through a series of events over several years The website at <http://nseus.org> documents materials from these events The international call for manuscripts for this volume requested the inclusion of major priorities and critical research areas methodological concerns and results of implementation of faculty professional development programs and reform in teaching in

undergraduate science classrooms In developing research manuscripts to be reviewed for RISE Volume 6 researchers were asked to consider the status and effectiveness of current and experimental practices for reforming undergraduate science courses involving all undergraduates including groups of students who are not always well represented in STEM education To influence practice it is important to understand how researchbased practice is made and how it is implemented The volume should be considered as a first step in thinking through what reform in undergraduate science teaching might look like and how we help faculty to implement such reform

Experimental Chemistry Russell A. Geanangel, Wesley William Wendlandt, 1979-08 *Working with Chemistry* Donald J. Wink, Sharon Fetzner-Gislason, Julie Ellefson Kuehn, 2004-02-20

With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well

Instructors Manual to Lab Manual Ralph Petrucci, William Harwood, Geoffrey Herring, 2001 *Exploring General Chemistry in the Laboratory* Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science

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will assess your knowledge and prepare you for competitive exams quizzes trivia and more

Chemistry: Molecules, Matter, and Change Media Activities Book Loretta Jones, Carl Hoeger, Peter William Atkins, Regina Schoenfield-Tacher, 2000-01-15 The Media Activity Book MAB for Jones Atkins Chemistry molecules matter and change contains chapters with lists and descriptions of some of the media available as you study the chapter Each activity begin with a specific textbook reference Then you are given a time estimate of how long it will take to use the media An M media icon in the margin of the textbook means that media exists to support that area of text The media is found in three different places on the website and on two CDs

Using Multimedia Technology in Chemistry Pre-laboratory Preparation Jeffrey Glen Yoder, 2002

Laboratory Experiments for General Chemistry Harold R. Hunt, Toby F. Block, George M. McKelvy, 2002 This established manual focuses on using non hazardous materials to teach the experimental nature of general chemistry Experiments are written to address students of various academic backgrounds and differing interests and abilities in chemistry While most experiments can be conducted in a single three hour period some have been designed to be completed over an extended time to illustrate that chemical systems do not work at an arbitrary schedule Suggestions are provided for combining experiments of shorter length and similar pedagogy

Chemical Investigations Nancy Konigsberg Kerner, 1986

Chemistry Frank Jenkins, 1992

Cracking the AP Chemistry Exam, 2020 Edition Princeton Review Staff, The Princeton Review, 2019-08-06 Cracking the AP Chemistry Exam 2020 Edition provides students with thorough subject reviews of all relevant topics including atomic structure thermodynamics the periodic table fundamental laws organic chemistry molecular binding and key equations laws and formulas It also includes helpful tables charts and diagrams and detailed advice on how to write a high scoring essay

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Green and Sustainable Chemistry and Engineering Concepcion Jimenez Gonzales, David J. C. Constable, 2025-09-17 The first textbook to fully integrate Green and Sustainable Chemistry and Engineering now in its second edition Green and Sustainable Chemistry and Engineering addresses key concepts and processes from an industrial and manufacturing perspective Using an integrated systems oriented approach this invaluable single volume resource bridges the divide between chemistry process design and engineering as well as environment health safety and life cycle considerations This revised new edition discusses trends in chemical processing that can lead to more sustainable practices explores new methods in the design of greener chemical synthesis addresses sustainability challenges and implementation issues and more Up to date examples and new practical exercises based on the broad experience of the

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Chemistry: The Easy Way
Joseph A. Mascetta, Mark Kernion, 2019-08-06 A self teaching guide for students Chemistry The Easy Way provides easy to follow lessons with comprehensive review and practice This edition features a brand new design and new content structure with illustrations and practice questions An essential resource for High school and college courses Virtual learning Learning pods Homeschooling Chemistry The Easy Way covers Atomic Structure Chemical Formulas Electrochemistry The Basics of Organic Chemistry And more

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